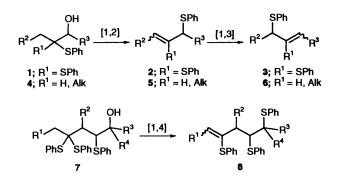
[1,4] Phenylthio Migrations in the Rearrangement of 2,4,4-Tris(phenylthio)butanols

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2,4,4-Tris(phenylthio)butanols rearranged by a [1,4]-PhS shift to give an *E,Z* mixture of vinyl sulfides with thionyl chloride but only the *Z* isomer with toluene-*p*-sulfonic acid. The [1,4]-PhS shift occurred only if another PhS group was present β to the hydroxyl group. The mechanism and scope of the reaction and the preparation of 2,4,4-tris(phenylthio)butanols *via* Michael addition of dithioacetal nucleophiles to methyl crotonate followed by α -sulfenylation is reported.

We know¹ that thionyl chloride and triethylamine cause [1,2] phenylthio (PhS) migration on 2,2-bis(phenylthio)ethanols 1 under conditions of kinetic control to give 2,3-bis(phenylthio)-propenes 2 and that on exposure to light isomerization of the allylic sulfides 2 and 3 occurs by a [1,3]-PhS shift. The alcohols

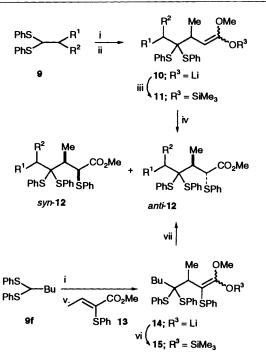


4 having only one PhS group have been transformed into the allyl sulfides 5 and 6 by similar acid-catalysed [1,2]- and photochemical [1,3]-PhS migrations on treatment with toluene-*p*-sulfonic acid (TsOH) under conditions of thermo-dynamic control.²

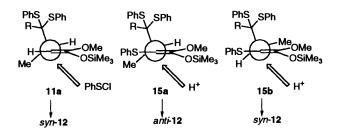
In a preliminary communication,³ we described the rearrangement of 2,4,4-tris(phenylthio)butanols 7, substrates with both bis(phenylthio) and phenylthio groups present in the same molecule. Under conditions of thermodynamic or kinetic control, an unusual⁴ long distance PhS shift, a [1,4]-PhS migration, gives the vinyl sulfides 8. In this paper, full details of the reaction are given including the extension of the [1,4]-PhS shift into substrates with a tertiary, secondary or primary migration terminus and the synthesis of the compounds used in this study.

Our route (Scheme 1) to prepare the starting material 2,4,4tris(phenylthio)butanols 7 started with a Michael addition of lithium derivatives of bis(phenylthio)acetals¹ 9 to methyl crotonate in THF at low temperature (-78 °C). The resulting lithium enolates 10 were silylated directly with chlorotrimethylsilane and triethylamine and the silyl enol ethers 11 were sulfenylated with benzenesulfenyl chloride in THF solution.⁵ Mixtures of diastereoisomeric methyl 2,4,4-tris(phenylthio)alkyl esters *syn*-12 and *anti*-12 were obtained in good yields, the *syn* isomer being the major product.[†] The results are summarized in Table 1.

The sense of the stereoselectivity is, therefore, that the bis(phenylthio)alkyllithium reagent and the PhS group add *anti* to the double bond of the ester and is the same as the tandem silyl cuprate–sulfenylation sequence we have already reported.⁶ It is not necessary to use copper with these bis(phenylthio)alkyl-



Scheme 1 Reagents and conditions: i, BuLi, TMEDA, THF, -78 °C; ii, methyl crotonate, THF, -78 °C; iii, Me₃SiCl, Et₃N, -78 °C to RT; iv, PhSCl, THF, -78 °C to RT; v, 13, THF, -78 °C to RT; vi, Me₃SiCl, Et₃N, -78 °C to RT; vii, AcOH, -78 °C to RT



lithium reagents. If the PhS group is already present in the unsaturated ester⁷ 13, conjugate addition of the lithiated bis(phenylthio) acetal 9d ($R^1 = Pr$, $R^2 = H$) followed by protonation of the silyl enol ether 14 with acetic acid

[†] The carbon chain is drawn in its most extended form: the compound is *syn* if PhS and Me are on the same side, and *anti* if they are on opposite sides. This is the convention introduced by Masamune. S. Masamune, T. Kaiho and D. S. Garvey, *J. Am. Chem. Soc.*, 1982, **104**, 5521; S. Masamune, W. Choy, J. S. Petersen and L. R. Sita, *J. Am. Chem. Soc.*, 1985, **24**, 1.

 Table 1
 Synthesis of methyl 2,4,4-tris(phenylthio)esters 12 and 2,4,4-tris(phenylthio)butanols 16

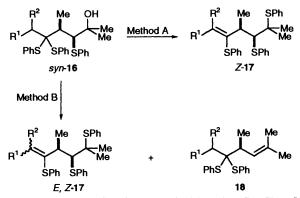
Starting material	R ¹	R ²	R ³	syn-12: anti-12ª	Yield(%) syn-12 ^b	Yield(%) syn-16°
9a	Н	н	Me	77:23	45 ^d	
9b	Me	Н	Me	78:22	40	64 (29)
9c	Me	Me	Me	е	43	78 Ì Ú
9d	Et	Н	Me	е	52	80
9e	CH ₂ =CH	Н	Me	86:14	60	42 (32)
9f	Pr	Н	Me	80:20	53	50 (39)
9g	PhCH,	Н	Me	65:40	65 ^d	40 (35)
9g	PhCH ₂	Н	Н	—	ſ	38 (35)

^{*a*} syn: anti Ratios were determined by ¹H NMR spectroscopy on the crude reaction mixture and stereochemistry established by coupling constants. ^{*b*} Isomers separated by flash column chromatography. ^{*c*} Yield of recovered syn-12 in brackets. ^{*d*} Yield of the mixture. ^{*e*} Ratio not determined. ^{*f*} syn-12h obtained in a 54% yield.

unexpectedly provided a 64:36 mixture of *syn*- and *anti*-esters **12** with weak *syn* selectivity.

The high *anti* diastereoselectivity of sulfenylation can be explained if the electrophile (PhSCl) attacks the less hindered face of the Houk conformation of the silyl enol ether **11a**.⁸ The bulk of the dithioacetal unit has been reported ⁹ to provide effective stereocontrol in alkylations of acyclic and cyclic α , β -unsaturated esters through the corresponding enolates. This fact, together with the lack of coordination by the sulfur group (it would have to form a seven-membered chelate) may also be responsible for the observed highly efficient * 1,2-asymmetric induction in alkylations of enolates.^{9a,10} On the other hand, the stereochemical outcome in the protonation of the silyl enol ether **15** can be rationalized if conformation **15b** is favoured for steric reasons. The electrophile (H⁺) attacks from the less hindered side to produce, as it happens, the same *syn*-isomer preferentially.

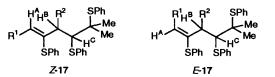
Methyllithium added cleanly to the 2,4,4-tris(phenylthio) esters syn-12 to give 2,4,4-tris(phenylthio)alcohols syn-16 (Table 1), which were submitted to thermodynamic and kinetic controlled [1,2]-PhS shift reaction conditions (Scheme 2).



Scheme 2 Reagents and conditions: Method A: TsOH, CH_2Cl_2 , reflux; Method B: $SOCl_2$, Et_3N , CCl_4 , 0 °C

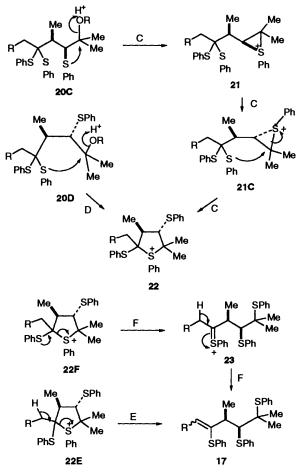
When the tertiary alcohols syn-16 were treated in the dark with TsOH in dichloromethane under reflux (Method A), reasonable yields of the [1,4]-PhS shift products, pure vinyl sulfides Z-17, were obtained (50–60%). Reaction of syn-16 with thionyl chloride and triethylamine in carbon tetrachloride at 0 °C for

a few minutes in the dark (Method B), provided a mixture of vinyl sulfides E-17 and Z-17 (50–65%), together with the alkene 18 (25–30%). The results are summarized in Table 2.



Configurations of the vinylic sulfides 17 were assigned on the basis that vinylic (H^A) and allylic (H^B) protons had different and characteristic signals in their NMR spectra, those for one isomer being outside those for the other. In fact, it has been reported ¹ that for 2,3-bis(phenylthio)propenes the vinyl proton resonance of the Z isomer occurs at lower field than that of E isomer, whereas the allylic proton resonance is at lower field in the E isomer than in the Z isomer. Table 3 gives a list of chemical shifts of vinylic (H^A) and allylic (H^B) protons for all compounds 17. The Z-isomer is more stable than the E-isomer, whose formation is favoured kinetically. The isomer ratios given in Table 2 were easily measured by NMR and give the composition of the mixture of vinyl sulfides E,Z-17 obtained by method B.

Mechanisms for [1,4]-PhS migration are summarized in Scheme 3 and we suggest a route *via* the five-membered ring intermediate 22. Formation of 22 could occur by initial formation 20C of an episulfonium ion 21, which would decompose by attack of the sulfur atom of the dithioacetal unit 21C, or by direct [1,4] participation by one PhS group of the bis(phenylthio) group with the loss of OR (route D, 20D). Route



Scheme 3 Mechanisms for the [1,4]-PhS shift

^{*} A lower diastereoselectivity has been reported in Fleming and Yamamoto's results on diastereoselective alkylation of enolates when using dimethylphenylsilyl and butyl, respectively, as stereocontrolling units.¹⁰

 Table 2
 Rearrangement of 2,4,4-tris(phenylthio)butanols 16

	R ¹ R ²			Method A Z-17 (%)	Method		
Starting material		R ²	R ³		17 (%)	E:Z	18 (%)
 syn-16b	Me	н	Me	55	52	57:48	25
syn-16c	Me	Me	Me	60	82		a
syn-16d	Et	Н	Me	64	68	71:29	a
syn-16e	CH ₂ =CH	н	Me	b	b	b	25
syn-16f	Pr	Н	Me	49	49	64:36	27
syn-16g	PhCH,	н	Me	47	52	69:31	34
syn-16h	PhCH ₂	Н	Н	с	53 d	22:78	a

^a 18c, 18d and 18h were not isolated. ^b Mixture of products: 17e not isolated. ^c 47% of a 73:27 mixture of Z: E-17h and 12% of a 63:37 mixture of Z: E-6-methyl-1-phenyl-3,5,6-trisphenylthiohept-3-ene 19h were obtained. ^d 8% of a 64:36 mixture of Z: E-19h was also obtained.

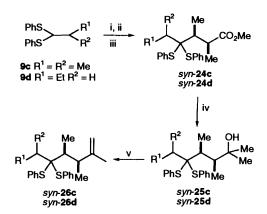
Table 3 Chemical shifts δ (p.p.m.) and coupling constants J (Hz) of the vinylic (H^A), allylic (H^B) and homoallylic (H^C) protons of the trisphenylthio compounds 17

Compd.	R ¹	R ²	H ^A	H ^B	H ^c
 Z-17b	Ме	Me	6.25	3.21	3.82
			q, J 6.6	q, J 5.9	d, J 1.6
<i>E-</i> 17b	Me	Me	5.49	3.93	3.75
			q, J 7.0	dq, J 2.5, 7.0	d, J 2.0
Z-17d	Et	Me	6.16	3.20	3.83
			t, J 7.0	q, J 7.0	S
<i>E</i> -17d	Et	Me	5.40	3.93	3.75
			t, J 7.3	dq, J 2.0, 7.0	d, J 2.0
Z-17f	Pr	Me	6.19	3.22	3.84
			t, J 6.6	q, J 6.9	d, J 1.5
<i>E</i> -17f	Pr	Me	5.45	3.93	3.75
			t, J 7.3	dq, J 2.4, 7.3	d, <i>J</i> 2.4
Z-17g	PhCH ₂	Me	6.34	3.29	3.88
			t, J 6.7	q, J 6.9	d, J 1.4
<i>E</i> -17g	$PhCH_2$	Me	5.59	4.10	3.82
			t, J 7.5	dq, J 2.2, 7.0	
<i>Z</i> -17h	$PhCH_2$	Н	6.31	m, 3.90-	-3.62
			t, J 7.3		
<i>E</i> -17h	PhCH ₂	Н	6.19	m, 3.90-	-3.62
			t, J 7.0		

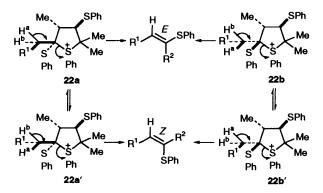
C can be rationalized on the basis that no [1,2]-PhS migration 'uphill' (secondary origin to tertiary terminus) via an episulfonium ion 21 has been observed without the assistance of silicon,^{2,12} while route D precludes the assistance of the PhS group β to the OH. The favourability of sulfur participation through cyclic sulfonium salts normally follows¹¹ the order of ring sizes: 3 > 5 > 6 > 4. Cleavage of the C-H bond can occur from 22 (route E) or from the rearranged cation 23 (route F). Formation of mixtures rich in the less stable *E*-17 using method B suggests that under these conditions route E should be preferred, as the transition state from 23 would so closely resemble the products 17 that Z-17 should be favoured. Under acidic conditions both routes may be possible.¹

Evidence against route D came from the by-product 18, formed by sulfur extrusion from 21. However, to test unequivocally the participation of the β -PhS group, tertiary alcohols lacking this group were prepared (Scheme 4). Thus, methyl 4,4-bis(phenylthio)esters syn-24c and syn-24d were available in good yields by 1,4-addition of lithium derivatives of bis(phenylthio)acetals 9c and 9d to methyl crotonate followed by methylation. Subsequent addition of 2 equiv. of methyllithium provided 4,4-bis(phenylthio)alcohols syn-25c and syn-25d, which were submitted to the rearrangement conditions. No [1,4]-PhS shift occurred, the dehydration product 26 alone being formed. We therefore prefer route C followed by E or F depending on the conditions.

The stereochemistry of the alkene resulting from the [1,4]-PhS shift can be rationalized (Scheme 5) on the basis of the



Scheme 4 Reagents and conditions: i, BuLi, TMEDA, THF, -78 °C; ii, methyl crotonate, THF, -78 °C; iii, MeI, -78 °C to RT; iv, MeLi, Et₂O, RT; v, Method B: SOCl₂, Et₃N, CCl₄, 0 °C



Scheme 5 Stereochemistry of double bond formation after the [1,4]-PhS shift

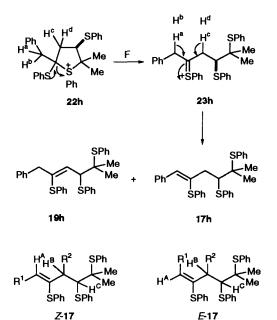
proposed mechanism. Either diastereotopic phenylthio group may participate in the conversion of 21 into 22a or 22b. Either C-H^a or C-H^b bonds in 22a and 22b can become *anti*-periplanar to the C-S⁺ bond being broken. The conformers 21a and 22b lead to the *E*-isomer of 17 whereas 22a' and 22b' lead to *Z*-17. The *Z*-isomer is thermodynamically favoured because the largest groups are *anti*. The kinetic preference for the *E* isomer follows from the relative instability of 22a', which has R¹ on the same surface of the molecule as the flanking PhS and Me groups. Method B therefore follows route E.

On the other hand, when method B was applied to 12h, mixtures rich in the more stable Z-isomer 17h were formed together with Z:E mixtures of 6-methyl-1-phenyl-3,5,6-tris-(phenylthio)hept-3-ene 19h. These facts suggest that mechanism F is operating in this case. The formation of 19h can only be explained if the loss of a proton (H^c or H^d) occurs from the rearranged cation 23h, whose geometry clearly favours the

 Table 4
 Synthesis of 2,4,4-trisphenylthioaldehydes 28

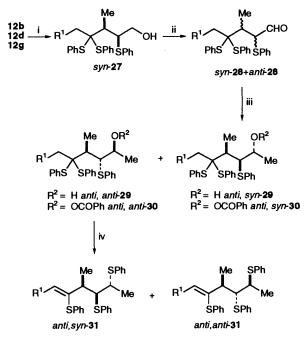
Starting material	R ¹	Oxidation method "	$T(^{\circ}\mathrm{C})/t^{b}$	syn -28: anti -28 °	Yield (%) of 28
syn-27b	Me	Swern (TEA)	- 78/15 min	89:11	70
syn-27d	Et	Swern (TEA)	$-78/15 \min$	82:18	64
syn-27d	Et	Swern (DIEA)	- 78/10 min	81:19	17
syn-27d	Et	Swern (DIEA)	-78/1 h	81:19	67
syn-27d	Et	Swern (TEA)	RT/48 h	41:58	_
syn-27d	Et	PDC	RT/24 h	81:19	30
syn-27d	Et	PCC	RT/3 h	63:37	40
syn-27d	Et	PCC	RT/1 h	66:34	45
syn-27g	PhCH ₂	Swern (TEA)	- 78/15 min	84:16	83

^a TEA = triethylamine; DIEA = diisopropylethylamine. ^b In Swern method, time and temperature of treatment with the base is given. ^c syn: anti Ratios were determined by ¹H NMR spectroscopy on the crude reaction mixture and stereochemistry established by coupling constants.



preferential formation of the Z-isomer. The same rearranged cation 23h could lose H^a or H^b and give Z-17h. Alternatively, formation of this conjugated alkene may be reversible.

To study the scope of the [1,4]-PhS shift, primary and secondary alcohols were also prepared (Scheme 6). Thus, the esters syn-12 were reduced with lithium aluminium hydride and primary alcohols syn-27 were obtained. However, all attempted rearrangements of these alcohols led to complex mixtures of products. The primary alcohols 27 were oxidized to the corresponding 2-phenylthio aldehydes 28. Subsequent addition of methyllithium provided secondary alcohols 29 in good yields. Oxidation of syn-27 was accomplished under Swern reaction conditions¹³ or by using PCC or PDC in dichoromethane solution.¹⁴ In all cases a mixture of diastereoisomeric aldehydes 28 was obtained, the syn isomer always being favoured (Table 4). Swern oxidation was performed at -78 °C only for 15 min as longer periods or higher temperatures favoured the loss of stereochemical integrity. We have reported the preparation of optically active 2-phenylthio aldehydes by a related route.¹⁵ The use of PCC or PDC gave slightly lower diastereoisomeric ratios but very poor yields (30-40%). This loss of stereochemical integrity is probably due to enolization of the 2-PhS aldehyde under the reaction conditions.¹⁵ Addition of methyllithium to



Scheme 6 Reagents and conditions: i, LiAlH₄, THF, 0 °C; ii, Swern oxidation; iii, MeLi Et₂O, RT; iv, SOCl₂, Et₃N, CCl₄, 0 °C

the aldehydes 28 provided an inseparable mixture of the two Felkin secondary alcohols anti, syn- and anti, anti-29.* The stereochemistry was assigned by NMR on the corresponding anti, syn- and anti, anti-benzoates 30, easily separated by flash column chromatography on silica gel. The 11:1 mixture of the alcohols 29 was rearranged with TsOH (method A) or thionyl chloride and triethylamine (method B). However, [1,4]-PhS shift occurred successfully only with method B when an E/Zmixture of a single diastereoisomer of the vinyl sufides E,Z-31was obtained in reasonable yield (55%). This single diastereoisomer can only be formed from the major diastereoisomer of 29 and so these experiments show that the reaction is stereospecific, but cannot definitely say in what sense. We assume that the [1,4]-PhS shift occurs with retention as inversion occurs in both the [1,2] and the [1,4] steps. Previously, secondary to secondary PhS shifts have been found with [1,4]-hydroxyl participation in the formation of tetrahydrofurans.¹⁶ Investigations into the application of the [1,4]-PhS shift are continuing: we have now found conditions where neither the PhS group next to the alcohol nor the second PhS group at the migration origin is necessary.¹⁷

Experimental

General experimental details have been published in this journal.¹⁸ Values of coupling constants (J) are quoted in Hz.

^{*} The syn and anti convention suggested by Masamune is used: the first syn or anti refers to the relationship of OH and PhS or PhS and PhS in the 2,3 positions and the second to the PhS and Me in the 3,4 positions when the chain is drawn in its extended form. See ref. 6.

Methyl(2RS, 3RS)-3-Methyl-2,4,4-tris(phenylthio)heptanoate 12d ($R^1 = Et, R^2 = H, R^3 = Me$).—*Typical procedure*. Butyllithium (1.6 mol dm⁻³ solution in hexane: 8.5 cm³, 13.6 mmol) was added to a stirred solution of 1,1-bis(phenylthio)butane 9d (3.16 g, 11.9 mmol) in dry THF (115 cm³) and TMEDA (1.8 cm³, 12.9 mmol) at 0 °C under argon. After 15 min, the solution was cooled to -78 °C and a solution of methyl crotonate (1.05 g, 10.5 mmol) in dry THF (10 cm³) was added dropwise. The mixture was stirred at this temperature for 1 h, after which trimethylsilyl chloride (6 cm³, 47 mmol) followed by triethylamine (6.6 cm³, 47 mmol) were added. The cooling bath was removed and the solution warmed to room temperature and stirred for 2 h. The contents of the flask were recooled to -78 °C and a solution of benzenesulfenyl chloride in THF was added dropwise. [This solution was prepared by addition of sulfuryl chloride (0.7 cm³, 8.7 mmol) to diphenyl disulfide (1.86 g, 8.5 mmol) in dry THF (10 cm³) containing dry pyridine (4 drops) and the mixture stirred for 2 h at room temperature under an argon atmosphere. The solution was used without further purification.] After 5 min, the mixture was allowed to come slowly to room temperature and then quenched with saturated aqueous ammonium chloride (150 cm³). The mixture was extracted with ether $(3 \times 100 \text{ cm}^3)$ and the combined extracts were washed with water $(2 \times 25 \text{ cm}^3)$, dried (MgSO₄), and evaporated under reduced pressure. The residue was purified by flash column chromatography on silica gel, eluting with hexane-dichloromethane (1:1), to give the ester syn-12d (2.63 g, 52%), as a viscous oil, $R_{\rm F}$ [hexane-CH₂Cl₂ (1:1)] 0.54; $v_{\rm max}({\rm film})/{\rm cm}^{-1}$ 1710 (CO) and 1570 (PhS); $\delta_{\rm H}({\rm CDCl}_3)$ 7.67– 7.18 (15 H, m, PhS), 4.68 (1 H, d, J 3.6, CHSPh), 3.56 (3 H, s, CO₂Me), 2.79 (1 H, dq, J 3.6 and 7, CHMe), 1.69-1.43 [4 H, m, (CH₂)₂], 1.39 (3 H, d, J7, CHMe) and 0.7 (3 H, t, J7, CH₂Me); $\delta_{\rm C}({\rm CDCl}_3)$ 14.0 (CHMe and CH₂Me), 18.1 (CH₂Me), 40.0 (CH₂CH₂Me), 42.6 (CHMe), 52.2 (CO₂Me), 53.6 (CHSPh), 74.5 [C(SPh)₂], 127.7, 128.1, 128.5, 128.6, 129.0, 129.3, 132.4, 137.0, 137.2 (ArC), 131.8, 132.0, 135.0 (ArC-S) and 174.4 (CO₂Me); m/z 343 (15%, M⁺ - C₇H₇OS), 233 (55, M⁺ -C₇H₇OS - PhSH), 123 (45, PhSCH₂), 110 (100, PhSH) and 109 (38, PhS) (Found: $M^+ - C_7H_7OS$, 343.1193. $C_{20}H_{23}OS_2$ requires $M - C_7 H_7 OS$, 343.1185).

Methyl (2RS,3RS)- and Methyl (2RS,3SR)-3-Methyl-2,4,4tris(phenylthio)pentanoate 12a ($R^1 = R^2 = H, R^3 = Me$).—In the same way, 1,1-bis(phenylthio)ethane 9a (2.46 g, 10 mmol), TMEDA (1.6 cm³, 11.2 mmol), butyllithium (1.6 mol dm⁻³ solution in hexane; 7.5 cm³, 12 mmol) in dry THF (100 cm³), methyl crotonate (0.9 g, 9 mmol) in dry THF (9 cm³), trimethylsilyl chloride (4.5 cm³, 35.5 mmol), triethylamine (5 cm³, 35.5 mmol) and benzenesulfenyl chloride [from sulfuryl chloride (0.6 cm³, 7.5 mmol), and diphenyl disulfide (1.57 g, 7.2 mmol) in dry THF (7 cm³)] gave an inseparable 3.5:1 mixture of the esters syn- and anti-12a (1.84 g, 45%) as an oil, $R_{\rm F}$ (hexane-dichloromethane, 1:1) 0.7; $\nu_{\rm max}$ (film)/cm⁻¹ 1720 (CO) and 1580 (PhS); $\delta_{\rm H}$ (CDCl₃) [*refers to syn isomer, **refers to anti isomer]: 7.66-7.17 (15 H, m, PhS), 4.84*, 4.43** (1 H, 2d, J 3.3 and 3.2, CHSPh), 3.71*, 3.58** (3 H, 2s, CO₂Me), 2.62*, 2.40** (1 H, 2 dq, J 3.3 and 6.9, and J 3.2 and 7.1, CHMe), 1.66**, 1.50* (3 H, 2d, J 7.1 and 6.9, CHMe), 1.32* and 1.27** [3 H, 2s, $Me(CSPh)_2$]; m/z 344 (3%, M⁺ – PhSH), 235 (50, M^+ – PhSH – PhS), 110 (100, PhSH) and 109 (43, PhS) (Found: $M^+ - C_6 H_6 S$, 344.0911. $C_{19} H_{20} O_2 S_2$ requires *M*-C₆H₆S, 344.090).

Methyl (2RS,3RS)- and Methyl (2RS,3RS)-3-Methyl-2,4,4tris(phenylthio)hexanoate **12b** ($R^1 = R^3 = Me$, $R^2 = H$).—In the same way, 1,1-bis(phenylthio)propane **9b** (6.5 g, 25 mmol), TMEDA (3.9 cm³, 28 mmol), butyllithium (1.6 mol dm⁻³ solution in hexane; 18.7 cm³, 30 mmol) in dry THF (150 cm³),

methyl crotonate (2.27 g, 22.7 mmol) in dry THF (20 cm³), trimethylsilyl chloride (13 cm³, 0.1 mol), triethylamine (14.5 cm³, 0.1 mol) and benzenesulfenyl chloride [from sulfuryl chloride (1.5 cm³, 18.6 mmol), and diphenyl disulfide (3.97 g, 18.2 mmol) in dry THF (20 cm³)] gave a 3.3:1 mixture of the esters syn- and anti-12b. Column chromatography on silica gel eluting with light petroleum (b.p. 30-40 °C)-ethyl acetate (18:1) gave the ester syn-12b (4.3 g, 40%), as an oil, $R_{\rm F}$ (hexanedichloromethane, 1:1) 0.42; $v_{max}(film)/cm^{-1}$ 1720 (CO) and 1560 (PhS); δ_H(CDCl₃) 7.64–7.18 (15 H, m, PhS), 4.68 (1 H, d, J 3.7, CHSPh), 4.57 (3 H, s, CO₂Me), 2.81 (1 H, dq, J 3.7 and 7, CHMe), 1.62 (2 H, m, CH₂Me), 1.40 (3 H, d, J7, CHMe) and 1.03 (3 H, t, J 7.2, CH₂Me); δ_{c} (CDCl₃): 9.5 (CH₂Me), 13.8 (CHMe), 30.7 (CH₂), 42.1 (CHMe), 52.2 (CO₂Me), 53.6 (CHSPh), 74.8 [C(SPh)₂], 127.6, 128.4, 128.5, 128.9, 129.2, 132.4, 137.2, 137.3 (ArC), 131.6, 134.9 (ArC-S) and 173.3 (CO_2Me) ; m/z 358 (6%, M⁺ – PhSH), 249 (100, M⁺ – PhSH – PhS), 217 (30, M⁺ – PhS – PhS – OMe), 189 (23, $M^+ - PhSH - PhS - C_2H_4O_2$, 181 (28, PhSCHCO₂Me), 149 (67, MeCHCSPh), 110 (95, PhSH) and 109 (42, PhS) (Found: $M^+ - C_6 H_6 S$, 358.1051. $C_{20} H_{22} O_2 S_2$ requires $M - C_6 H_6 S_2$ C₆H₆S, 358.1056).

Methyl (2R,3RS)-3,5-Dimethyl-2,4,4-tris(phenylthio)hexanoate 12c ($R^1 = R^2 = R^3 = Me$).—In the same way, 2-methyl-1,1bis(phenylthio)propane 9c (1.13 g, 4.1 mmol), TMEDA (0.59 cm³, 4.6 mmol), butyllithium (1.6 mol dm⁻³ solution in hexane; 3 cm³, 4.8 mmol) in dry THF (40 cm³), methyl crotonate (0.38 g, 3.8 mmol) in dry THF (4 cm³), trimethylsilyl chloride (2.3 cm³, 18 mmol), triethylamine (2.5 cm³, 18 mmol) and benzenesulfenyl chloride [from sulfuryl chloride (0.25 cm³, 3.1 mmol), and diphenyl disulfide (0.66 g, 3.05 mmol) in dry THF (5 cm³)] gave the ester syn-12c (0.78 g, 43%) as an oil, $R_{\rm F}$ [hexanedichloromethane (1:1)] 0.55; $v_{max}(film)/cm^{-1}$ 1720 (CO) and 1580 (PhS); $\delta_{\rm H}$ (CDCl₃) 7.73–7.19 (15 H, m, PhS), 4.87 (1 H, d, J 3.2, CHSPh), 3.53 (3 H, s, CO₂Me), 3.00 (1 H, dq, J 3.2 and 7.2, CHMe), 2.46 (1 H, septuplet, J 6.5, CHMe₂), 1.24 (3 H, d, J 7.2, CHMe) and 1.17 (6 H, 2 d, J 6.5, CHMe₂); δ_{C} (CDCl₃) 14.4 (CHMe), 17.4, 20.8 (HCMe₂), 36.1 (HCMe₂), 43.5 (CHMe), 52.0 (CO₂Me), 54.5 (CHSPh), 78.2 [C(SPh)₂], 127.6, 128.5, 128.6, 128.8, 129.0, 129.1, 132.4, 136.2, 137.6 (ArC), 131.6, 133.2, 135.3 (ArC-S) and 173.7 (CO₂Me); m/z 372 (14%, $PhSH - PhS - C_2H_4O_2$, 191 (20, M⁺ - PhSCH₂CO₂Me), 182 (18, PhSCH₂CO₂Me), 149 (30, MeCHCSPh) 123 (17, PhSCH₂), 110 (100, PhSH) and 109 (35, PhS) (Found: M⁺ - C_6H_6S , 372.1232. $C_{21}H_{24}O_2S_2$ requires $M - C_6H_6S$, 372.1212).

Methyl (2RS,3RS)-3-Methyl-2,4,4-tris(phenylthio)hept-6enoate 12e ($R^1 = CH = CH_2$, $R^2 = H$, $R^3 = Me$).—In the same way, 1,1-bisphenylthiobut-3-ene 9e (1.03 g, 3.76 mmol), TMEDA (0.675 cm³, 5.3 mmol), butyllithium (1.6 mol dm⁻³ solution in hexane; 3 cm, 4.8 mmol) in dry THF (40 cm³), methyl crotonate (0.342 g, 3.4 mmol) in dry THF (4 cm³), trimethylsilyl chloride (2.3 cm³, 18 mmol), triethylamine (2.5 cm³, 18 mmol) and benzenesulfenyl chloride [from sulfuryl chloride (0.23 cm³, 2.9 mmol), and diphenyl disulfide (0.59 g, 2.7 mmol) in dry THF (5 cm³)] gave the ester syn-12e (0.98 g, 60%) as an oil, $R_{\rm F}$ [hexane-dichloromethane (1:1)] 0.35; $v_{max}(film)/cm^{-1}$ 1720 (CO) and 1580 (PhS); $\delta_{H}(CDCl_3)$ 7.62– 7.20 (15 H, m, PhS), 5.96 (1 H, m, HC=CH₂), 5.00 (2 H, m, HC=CH₂), 4.73 (1 H, d, J 3.4, CHSPh), 3.58 (3 H, s, CO₂Me), 2.80 (1 H, dq, J 3.4 and 6.9, CHMe), 2.35 (2 H, m, H_2 CHC=CH₂) and 1.43 (3 H, d, J 7.2, CHMe); δ_c (CDCl₃) 13.7 (CHMe), 42.5 (CH₂), 42.8 (CHMe), 52.2 (CO₂Me), 53.6 (CHSPh), 72.5 [C(SPh)₂], 118.1 (CH=CH₂), 127.6, 128.5, 128.6, 128.9, 129.2, 129.3, 129.4, 132.4, 132.9, 133.5, 137.3, 137.4,

137.5, 137.6 (ArC, CH=CH₂), 131.5, 135.0 (ArC-S) and 173.3 (CO₂Me); m/z 371 (7%, M⁺ – PhS), 261 (40, M⁺ – PhSH – PhS), 189 (68, M⁺ – PhS – PhSCH₂CO₂Me), 182 (20, PhSCH₂Me), 149 (23, MeCHCSPh), 123 (23, PhSCH₂), 110 (100, PhSH) and 109 (40, PhS) (Found: M⁺ – C₆H₅S, 371.1155. C₂₁H₂₃O₂S₂ requires $M - C_6H_5S$, 371.1134).

Methyl (2RS,3RS)- and Methyl (2RS,3SR)-3-Methyl-2,4,4tris(phenylthio)heptanoate 12f ($R^1 = Pr, R^2 = H, R^3 = Me$). -In the same way, 1,1-bis-phenylthiopentane 9f (3.9 g, 13.5 mmol), TMEDA (2.1 cm³, 15 mmol), butyllithium (1.6 mol dm⁻³ solution in hexane; 10 cm³, 16 mmol) in dry THF (135 cm³), methyl crotonate (1.23 g, 12.3 mmol) in dry THF (12 cm³), trimethylsilyl chloride (7 cm³, 55 mmol), triethylamine (8 cm³, 55 mmol) and benzenesulfenyl chloride [from sulfuryl chloride (0.8 cm³, 10 mmol), and diphenyl disulfide (2.14 g, 9.8 mmol) in dry THF (10 cm³)] gave the ester syn-12f (3.2 g, 53%) as an oil, $R_{\rm F}$ [hexane-dichloromethane (1:1)] 0.46; $v_{\rm max}({\rm film})/{\rm cm}^{-1}$ 1720 (CO) and 1580 (PhS); $\delta_{\rm H}({\rm CDCl}_3)$ 7.64– 7.20 (15 H, m, PhS), 4.69 (1 H, d, J 3.6, CHSPh), 3.57 (3 H, s, CO₂Me), 2.80 (1 H, dq, J 3.6 and 7.2, CHMe), 1.48-0.80 [6 H, m, (CH₂)₃Me], 1.63 (3 H, d, J7, CHMe) and 0.73 (3 H, t, J7.2, CH₂Me); $\delta_{\rm C}$ (CDCl₃) 13.8 (CHMe and CH₂Me), 22.1 (CH_2Me) , 26.6 (CH_2CH_2Me) , 37.6 $[CH_2(CH_2)_2Me]$, 42.5 (CHMe), 52.1 (CO₂Me), 53.5 (CHSPh), 74.3 [C(SPh)₂], 127.5, 128.4, 128.5, 128.9, 129.1, 129.2, 129.6, 132.9, 133.5, 135.6, 136.9, 137.2, 137.2 (ArC), 131.7, 131.9, 134.8 (ArC-S) and 173.3 $(CO_2Me); m/z$ 386 $(27\%, M^+ - PhSH), 277$ (100, M⁺ PhSH - PhS), 245 (30, $M^+ - PhS - PhS - OMe$), 217 (23, $M^+ - PhS - PhSH - C_2H_4O_2$, 205 [20, Me(CH₂)₃CSPh-CHMe], 181 (25, PhSCHCO₂Me), 149 (30, MeCHCSPh), 121 (40, PhSCH₂), 110 (PhSH) and 109 (47, PhS) (Found: M^+ – C_6H_6S , 386.1374. $C_{22}H_{26}O_2S_2$ requires $M - C_6H_6S_2$ 386.1368).

Methyl (2RS,3RS)- and Methyl (2RS,3SR)-3-Methyl-6phenyl-2,4,4-tris(phenylthio)hexanoate 12g ($R^1 = CH_2Ph$, $R^2 = H$, $R^3 = Me$).—In the same way, 3-phenyl-1,1-bisphenylthiopropane 9g (8.4 g, 25 mmol), TMEDA (3.9 cm³, 28 mmol), butyllithium (1.6 mol dm⁻³ solution in hexane; 19 cm³, 30 mmol) in dry THF (150 cm³), methyl crotonate (2.3 g, 22.3 mmol) in dry THF (20 cm), trimethylsilyl chloride (13 cm³, 0.1 mol), triethylamine (14.5 cm³, 0.1 mol) and benzenesulfenyl chloride [from sulfuryl chloride (1.5 cm³, 18.6 mmol), and diphenyl disulfide (3.97 g, 18.2 mmol) in dry THF (20 cm³)] gave the esters syn and anti-12g (8.1 g, 65%), identified by their spectroscopic data [* refers to syn isomer; ** refers to anti isomer]: $v_{max}(film)/cm^{-1}$ 1720 (CO) and 1580 (PhS); $\delta_{H}(CDCl_{3})$ 7.75-7.09 (20 H, m, PhS and Ph), 4.77*, 4.60** (1 H, 2d, J 3.5 and 2.8, CHSPh), 3.72**, 3.57* (3 H, 2s, CO₂Me), 2.98-2.61 (3 H, m, CH₂CH₂Ph and CHMe), 2.19–1.58 (2 H, m, CH₂Ph) and 1.54**, 1.48* (3 H, 2d, J 7.2 and 7, CHMe); δ_C(CDCl₃): 11.8**, 13.9* (CHMe), 30.3**, 30.9* (CH₂Ph), 39.3**, 39.5* (CH₂CH₂Ph), 42.6*, 47.2** (CHMe), 52.0**, 52.1* (CO₂Me), 52.6**, 53.4* (CHSPh), 72.1**, 73.8* [C(SPh)₂], 125.8, 127.6, 128.1, 128.3, 128.5, 128.7, 128.9, 129.1, 129.2, 132.4, 132.9, 136.7, 136.9, 137.1, 137.2 (ArC), 131.7, 134.8, 141.3 (ArC-S), ArC-C) and 172.2**, 173.3* (CO₂Me); m/z 435 (1%, M⁺ PhSH), 325 (50, M⁺ – PhSH – PhSH), 215 (20, M⁺ – PhSH – PhSH – PhSH), 181 (19, PhSCHCO₂Me), 123 (27, PhSCH₂), 110 (70, PhSH), 109 (27, PhS) and 91 (100, PhCH₂) (Found: $M^+ - C_6H_6S$, 435.1455. $C_{26}H_{27}O_2S_2$ requires $M - C_6 H_6 S$, 435.1446). These isomers could be separated by flash column chromatography on silica eluting with light petroleum (b.p. 30-40 °C)-ethyl acetate, 20:1.

Methyl 6-Phenyl-2,4,4-tris(phenylthio)hexanoate 12h ($R^1 = CH_2Ph$, $R^2 = H$, $R^3 = Me$).—In the same way, 3-phenyl-1,1-

bis(phenylthio)propane 9g (1.7 g, 5.06 mmol), TMEDA (0.8 cm^3 , 5.7 mmol), butyllithium (1.6 mol dm⁻³ solution in hexane; 4.2 cm^3 , 6.7 mmol) in dry THF (50 cm^3), methyl acrylate (0.46 g, 4.6 mmol) in dry THF (5 cm³), trimethylsilyl chloride (3 cm³, 23 mmol), triethylamine (3.5 cm³, 23 mmol) and benzene sulfenyl chloride [from sulfuryl chloride (0.32 cm³, 3.9 mmol), and diphenyl disulfide (0.83 g, 3.81 mmol) in dry THF (10 cm³)] gave the ester 12h (1.35 g, 54%) as an oil, $R_{\rm F}$ (hexanedichloromethane, 1:1) 0.53; $v_{max}(film)/cm^{-1}$ 1720 (CO) and 1580 (PhS); $\delta_{\rm H}$ (CDCl₃) 7.71–6.92 (20 H, m, PhS and Ph), 4.36 (1 H, dd, J 2.7 and 8.5, CHSPh), 3.54 (3 H, s, CO₂Me), 3.02-2.83 (2 H, m, CH₂Ph), 2.59 (1 H, dd, J 8.5 and 15, CH_AH_BCHSPh), 2.10 (1 H, dd, J 2.7 and 15, CH_AH_BCHSPh) and 1.85 (2 H, t, J 8, CH₂CH₂Ph); δ_C(CDCl₃) 30.8 (PhCH₂), 39.7, 40.4 (CH₂CH₂Ph and CH₂CHSPh), 46.6 (CHSPh), 52.3 (CO₂Me), 68.1 [C(SPh)₂], 125.9, 128.3, 128.4, 128.6, 128.8, 128.9, 129.2, 129.5, 133.9, 136.6, 136.7 (ArC), 130.6, 131.0, 132.5, 141.1 (ArC-S, ArC-C) and 172.8 (CO₂Me); m/z 421 (2%, M⁺ – PhS), 311 (4, M^+ – PhS – PhSH), 201 (2, M^+ – PhS – PhSH – PhSH), 110 (27, PhSH) and 91 (100, PhCH₂) (Found: $M^+ - C_6H_5S$, 421.1296. $C_{25}H_{25}O_2S_2$ requires $M - C_6H_5S$, 421.1290).

Conjugate Addition of 1,1-Bis(phenylthio)butane 9d to Methyl α -Phenylthiobut-2-enoate 13.—Butyllithium (1.6 mol dm⁻³ solution in hexane; 3.5 cm³, 5.6 mmol) was added to a stirred solution of 9d (1.38 g, 5.2 mmol) in dry THF (35 cm³) and TMEDA (0.7 cm³, 5 mmol) at 0 °C under argon. After 15 min, the solution was cooled to -78 °C and a solution of the 2-phenylthioacrylate 13 (0.95 g, 4.8 mmol) in dry THF (5 cm³) was added dropwise. The mixture was stirred at this temperature for 15 min, after which acetic acid (0.5 cm³) in THF (1 cm³) was added. The solution was left to warm to room temperature and then saturated aqueous ammonium chloride (50 cm³) was added. The mixture was extracted with ether $(3 \times 50 \text{ cm}^3)$ and the combined extracts were washed with water $(2 \times 15 \text{ cm})$, dried (MgSO₄) and evaporated under reduced pressure. The residue was purified by flash column chromatography on silica gel, eluting with hexane-dichloromethane (1:1), to give a 1.8:1 mixture of esters syn- and anti-12d (1.13 g, 45%).

(3RS,4RS)-2,4-Dimethyl-3,5,5-tris(phenylthio)heptan-2-ol **16b**: *Typical Procedure*.—Methyllithium (1.4 mol dm⁻³solution in ether; 6 cm³, 8.4 mmol) was syringed into a flask, equipped with a magnetic stirrer, maintained under an argon atmosphere and containing dry ether (30 cm³). The mixture was cooled to 0 °C and solution of the syn-ester 12b (1.7 g, 3.8 mmol) in ether (8 cm³) was slowly added. The mixture was stirred for 1 h after which it was hydrolysed with saturated aqueous ammonium chloride (40 cm³) and extracted with ether (3 \times 30 cm^3). The combined extracts were dried (MgSO₄) and evaporated under reduced pressure. Purification of the residue by flash column chromatography on silica gel, eluting with hexane-dichloromethane (1:1), gave recovered syn-ester 12b (0.52 g, 29%) and the alcohol syn-16b (1.15 g, 91% based on recovered starting material) as an oil $R_{\rm F}$ (CH₂Cl₂-hexane, 2.5:1) 0.47; $v_{max}(film)/cm^{-1}$ 3650–3300 (OH) and 1600 (PhS); $\delta_{\rm H}({\rm CDCl}_3)$ 7.73–7.10 (15 H, m, PhS), 4.26 (1 H, s, CHSPh), 2.66 (1 H, q, J 6.9, CHMe), 2.62–2.40 (1 H, br s, OH), 1.69–1.57 (1 H, m, CH_AH_BMe), 1.43–1.37 (1 H, m, CH_AH_BMe), 1.35, 1.23 (6 H, 2s, HOCMe₂), 1.27 (3 H, d, J 6.9, CHMe) and 1.03 (3 H, t, J 7.1, CH₂Me): δ_c(CDCl₃) 9.3 (CH₂Me), 14.3 (CHMe), 27.3, 27.9 (HOCMe₂), 29.9 (CH₂), 40.2 (CHMe), 61.2 (CHSPh), 75.1 (COH), 75.4 [C(SPh)₂], 125.6, 128.1, 128.4, 128.6, 128.8, 129.0, 129.2, 137.3, 137.6 (ArC), 131.6, 132.6 and 138.4 (ArC-S); *m*/*z* 358 (83%, M^+ – PhSH), 300 [63, MeCH₂C(SPh)₂CHMe-CH⁺], 249 (51, M⁺ – PhSH – PhS), 232 (12, M⁺ – PhSH – PhS – OH), 191 (100, M⁺ – PhS – PhS –

 C_3H_7O), 181 (37, Me₂COHCHSPh), 163 (25, Me₂C=CPhS), 149 (23, MeCH=CSPh), 139 (34, M⁺ – PhS – PhSH – PhSH), 123 (55, PhSCH₂) and 110 (21, PhSH) (Found: M⁺ – C_6H_6S , 318.1432, $C_{21}H_{26}OS_2$ requires $M - C_6H_6S$, 318.1419).

(3RS,4RS)-2,4,6-Trimethyl-3,5,5-tris(phenylthio)heptan-2-ol 16c.—In the same way, the syn-ester 12c (0.365 g, 0.75 mmol) and methyllithium (1.5 mol dm⁻³ solution in ether; 1.2 cm³, 1.8 mmol) in dry ether (8 cm³) gave the alcohol syn-16c (0.285 g, 78%) as an oil, $R_{\rm F}$ (dichloromethane-hexane, 2.5:1) 0.52; $v_{max}(film)/cm^{-1}$ 3500–3350 (OH) and 1580 (PhS); $\delta_{H}(CDCl_{3})$ 7.73-7.13 (15 H, m, PhS), 4.41 (1 H, s, CHSPh), 2.92 (1 H, br s, OH), 2.84 (1 H, q, J7, CHMe), 2.47 (1 H, 2q, J 6.5, CHMe₂), 1.36, 1.20 (6 H, 2d, J 6.5, CHMe2), 1.31, 1.21 (6 H, 2s, HOC- Me_2) and 1.09 (3 H, d, J 7, CHMe); δ_C (CDCl₃) 14.7 (CHMe), 18.5, 21.2 (CHMe₂), 26.2, 29.0 (HOCMe₂), 35.0 (CHMe₂), 42.8 (CHMe), 61.5 (CHSPh), 75.9 [COH, C(SPh)2], 125.7, 128.3, 128.4, 128.8, 129.1, 129.3, 131.4, 135.8, 138.2 (ArC), 134.0 and 138.9 (ArC-S); m/z 372 (1 %, M⁺ – PhSH), 263 (33, M⁺ – PhSH – PhS), 262 (33, M⁺ – PhSH – PhSH), 205 $(15, M^+ - PhS - PhS - C_3H_7O), 153 (13, M^+ - PhS - PhS$ PhSH - PhSH), 110 (100, PhSH) and 109 (28, PhS) (Found: $M^+ - C_6 H_6 S$, 372.1571. $C_{22} H_{28} OS_2$ requires $M - C_6 H_6 S$, 372.1575).

(3RS,4RS)-2,4-Dimethyl-3,5,5-tris(phenylthio)octan-2-ol**16d**.—In the same way, the syn-ester **12d** (0.23 g, 0.47 mmol) and methyllithium (1.5 mol dm⁻³ solution in ether; 0.7 cm³, 1.05 mmol) in dry ether (5 cm³) gave the alcohol syn-**16d** (0.185 g, 80%) as an oil, identified by its ¹H NMR spectrum, R_F (dichloromethane-hexane, 2.5:1) 0.53; $v_{max}(film)/cm^{-1}$ 3500-3300 (OH) and 1580 (PhS); $\delta_H(CDCl_3)$ 7.73-7.14 (15 H, m, PhS), 4.26 (1 H, s, CHSPh), 2.67 (1 H, q, J6.5, CHMe), 2.31 (1 H, br s, OH), 1.82-1.55 (2 H, m, CH₂CH₂Me), 1.53-1.20 (2 H, m, CH₂CH₂Me), 1.34, 1.22 (6 H, 2 s, HOCMe₂), 1.28 (3 H, d, J6.8, CHMe) and 0.66 [3 H, t, J 7.1, (CH₂)₂Me].

(3RS,4RS)-2,4-Dimethyl-3,5,5-tris(phenylthio)-oct-7-en-2-ol 12e.—In the same way, the syn-ester 12e (0.825 g, 1.71 mmol) and methyllithium (1.4 mol dm⁻³ solution in ether; 2.7 cm³, 3.78 mmol) in dry ether (10 cm^3) gave the alcohol syn-16e (0.350 g,42%) as an oil, R_F [dichloromethane-hexane (2.5:1)] 0.52; $v_{\rm max}({\rm film})/{\rm cm}^{-1}$ 3500–3300 (OH) and 1580 (PhS); $\delta_{\rm H}({\rm CDCl}_3)$ 7.73-7.13 (15 H, m, PhS), 6.11-6.00 (1 H, m, HC=CH₂), 5.07-5.00 (2 H, m, HC=CH₂), 4.29 (1 H, s, CHSPh), 2.63 (1 H, q, J6.8, CHMe), 2.58 (1 H, br s, OH), 2.43-2.20 (2 H, m, CH₂CH=CH₂), 1.34, 1.22 (6 H, 2 s, HOCMe₂) and 1.31 (3 H, d, J 6.8, CHMe); δ_c(CDCl₃) 14.3 (CHMe), 27.4, 27.8 (HOCMe₂), 41.1 (CHMe), 41.6 (CH₂CH=CH₂), 61.3 (CHSPh), 75.1 (COH), 75.3 [C(SPh)₂], 117.9 (HC=CH₂), 125.7, 128.3, 128.5, 128.9, 129.2, 129.4, 133.6, 137.4, 137.7 (ArC, HC=CH₂), 131.5, 132.2 and 138.3 (ArC-S) m/z 370 (5%, M⁺ – PhSH), 261 (35, M⁺ PhSH – PhS), 189 (20, M^+ – PhSH – PhSCHCMe₂OH), 181 (26, Me₂COHCHSPh), 123 (26, PhSCH₂), 110 (100, PhSH) and 109 (32, PhS) (Found: $M^+ - C_6H_6S$, 370.1443. $C_{22}H_{26}OS_2$ requires $M - C_6H_6S$, 370.1419).

(3RS,4RS)-2,4-Dimethyl-3,5,5-tris(phenylthio)nonan-2-ol

16f.—In the same way, the *syn*-ester **12f** (0.9 g, 1.8 mmol) and methyllithium (1.4 mol dm⁻³ solution in ether; 3 cm³, 4.2 mmol) in dry ether (20 cm³) gave recovered *syn*-ester (0.350 g, 39%); and the *alcohol syn*-**16f** (0.45 g, 82% based on recovered starting material) as an oil $R_{\rm F}$ [dichloromethane–hexane (2.5:1)] 0.54; $\nu_{\rm max}$ (film)/cm⁻¹ 3660–3200 (OH) and 1570 (PhS); $\delta_{\rm H}$ (CDCl₃) 7.72–7.13 (15 H, m, PhS), 4.26 (1 H, s, CHSPh), 1.44 (1 H, q, J 6.9, CHMe), 2.61–2.50 (1 H, br s, OH), 1.70–1.37 [2 H, m, CH₂(CH₂)₂Me], 1.34, 1.21 (6 H, 2 s, HOCMe₂), 1.34–1.21 (2 H, m, CH₂CH₂CH₂Me), 1.27 (3 H, d, J6.9, CHMe), 1.03–0.95 (2 H, m, CH_2Me) and 0.73 (3 H, t, J 7.2, CH_2Me); $\delta_C(CDCl_3)$: 8.8 (CH_2Me), 9.4 (CHMe), 17.7 (CH_2Me), 21.5 (CH_2CH_2Me), 22.2, 22.8 (HOC Me_2), 32.0 [$CH_2(CH_2)_2Me$], 35.9 (CHMe), 56.3 (CHSPh), 69.9 (COH), 70.1 [$C(SPh)_2$], 120.6, 123.1, 123.4, 123.6, 123.8, 123.9, 124.1, 132.1, 132.5 (ArC), 126.7, 127.7 and 133.3 (ArC-S); m/z 386 (3%, M⁺ – PhSH), 277 (8, M⁺ – PhSH – PhS), 219 (30, M⁺ – PhS – $PhS - C_3H_7O$), 181 (10, Me₂COHCHSPh), 163 (20, Me₂C=CPhS), 123 (37, PhSCH₂), 110 (100, PhSH) and 109 (32, PhS) (Found: M⁺ – C_6H_6S , 386.1761. $C_{23}H_{30}OS_2$ requires $M - C_6H_6S$, 386.1713).

(3RS,4RS)-2,4-Dimethyl-7-phenyl-3,5,5-tris(phenylthio)heptan-2-ol 16g.—In the same way, the syn-ester 12g (3.26 g, 6 mmol) and methyllithium (1.4 mol dm⁻³ solution in ether; 10 cm³, 14 mmol) in dry ether (60 cm³) gave recovered syn-ester 12g(1.14 g, 35%); and the *alcohol syn*-16g(1.30 g, 62%) as an oil $R_{\rm F}$ [dichloromethane-hexane (2.5:1)] 0.50; $v_{\rm max}({\rm film})/{\rm cm}^{-1}$ 3660–3240 (OH) and 1600 (PhS); $\delta_{\rm H}$ (CDCl₃) 7.78–6.80 (20 H, m, PhS and Ph), 4.32 (1 H, s, CHSPh), 2.96 (1 H, dt, J4 and 13.5, CH₂CH_AH_BPh), 2.79 (1 H, q, J 6.8, CHMe), 2.68 (1 H, dt, J 4.9 and 13, CH_AH_BCH₂Ph), 2.63 (1 H, br s, OH), 2.09 (1 H, dt, J4.9 and 14.5, CH_AH_BCH₂Ph), 1.81 (1 H, dt, J 4 and 12.7, $CH_2CH_AH_BPh$) and 1.36 and 1.24 (6 H, 2 s, $HOCMe_2$); $\delta_{\rm C}({\rm CDCl}_3)$: 14.5 (CHMe), 27.3, 28.1 (HOCMe₂), 30.8 (CH₂-CH₂Ph), 38.8 (CH₂CH₂Ph), 41.1 (CHMe), 61.2 (CHSPh), 74.5 (COH), 75.3 [*C*(SPh)₂], 125.7, 125.8, 128.1, 128.3, 128.5, 128.7, 128.9, 129.0, 129.1, 129.3, 137.0 (ArC), 131.8, 132.6, 138.2 and 141.3 (ArC-S, ArC-C); m/z 434 (4%, M⁺ – PhSH), 417 (3, M⁺ - PhSH - OH), 375 [2, M⁺ - PhSH - C₃H₂O], 343 (1, $M^+ - PhSH - PhCH_2$), 325 (13, $M^+ - PhSH - PhS$), 307 $(15, M^+ - PhSH - PhSH - OH), 267 (44, M^+ - PhS - P$ $PhS - C_{3}H_{7}O)$, 215 (12, $M^{+} - PhSH - PhSH - PhS)$, 181 (19, Me₂COHCHSPh), 163 (19, Me₂C=CPhS), 157 (22, M⁺ $PhS - PhS - PhS - C_3H_7O$), 143 (15, $M^+ - PhSH - Ph-$ SH – Me₂COHCHSPh), 123 (57, PhSCH₂), 110 (75, PhSH) and 91 (100, PhCH₂) (Found: $M^+ - C_6H_6S$, 434.1721. $C_{27}H_{30}OS_2$ requires $M - C_6H_6S$, 434.1713).

2-Methyl-7-phenyl-3,5,5-tris(phenylthio)heptan-2-ol 16h.—In the same way, the syn-ester 12h (0.86 g, 1.62 mmol) and methyllithium (1.4 mol dm⁻³ solution in ether; 2.5 cm³, 3.6 mmol) in dry ether (15 cm³) gave recovered syn-ester 12h (0.3 g, 35%); and the alcohol syn-16h (0.33 g, 59%) as an oil $R_{\rm F}$ [dichloromethane-hexane (2.5:1)] 0.50; $v_{max}(film)/cm^{-1}$ 3640-3200 (OH) and 1600 (PhS); $\delta_{\rm H}(\rm CDCl_3)$ 7.85–6.93 (20 H, m, PhS and Ph), 3.79 (1 H, dd, J 1.7 and 8.1, CHSPh), 3.14 (1 H, dt, J 4.8 and 13, CH₂CH_ACH_BPh), 2.98 (1 H, dt, J 4.2 and 13, $CH_2CH_AH_BPh$), 2.40–2.16 (4 H, m, CH_2CHSPh , CH_A - H_BCH_2Ph and OH), 1.84 (1 H, dt, J4.2 and 13, $CH_AH_BCH_2Ph$) and 1.11 and 1.05 (6 H, 2 s, HOCMe2); $\delta_{\rm C}({\rm CDCl}_3)$ 25.5, 26.9 (HOCMe₂), 30.8 (CH₂CH₂Ph), 38.7, 39.4 (CH₂CH₂Ph and CH₂CHSPh), 58.0 (CHSPh), 69.6 (COH), 73.9 [C(SPh)₂], 125.7, 126.1, 128.2, 128.4, 128.7, 128.8, 129.0, 129.4, 134.9, 136.5 (ArC), 131.5, 132.4, 137.5 and 141.41 (ArC-S, ArC-C); m/z 420 $(2\%, M^+ - PhSH), 403 (1, M^+ - PhSH - OH), 311 (96, M^+)$ - PhSH - PhS), 293 (19, M⁺ - PhSH - PhSH - OH), 201 (12, M⁺ - PhSH - PhSH - PhSH - PhS), 143 (24, M⁺ -PhSH – PhSH – Me₂COHCHSPh), 123 (55, PhSCH₂), 110 (100, PhSH) and 91 (91, PhCH₂) (Found: $M^+ - C_6H_6S$, 420.1518. $C_{26}H_{28}OS_2$ requires $M - C_6H_6S$, 420.1575).

Rearrangement of (3RS,4RS)-2,4-Dimethyl-3,5,5-tris(phenylthio)heptan-2-ol **16b** by Method A: Typical Procedure.—The syn-alcohol **16b** (128 mg, 0.27 mmol) was refluxed in dry dichloromethane (1.2 cm³) with toluene-p-sulfonic acid (TsOH) (12 mg, 0.07 mmol) for 15 min. The solution was cooled, poured into saturated aqueous sodium carbonate (3 cm³) and the

mixture extracted with dichloromethane $(3 \times 5 \text{ cm}^3)$. The combined extracts were dried (MgSO₄) and evaporated under reduced pressure. Purification of the residue by flash column chromatography on silica gel eluting with a mixture of hexanedichloromethane (3:1) gave the vinyl sulfide (Z)-(4RS,5RS)-4,6-dimethyl-3,5,6-tris(phenylthio)hept-2-ene 17b (67 mg, 55%), as an oil R_F [hexane-dichloromethane (3:1)] 0.46; v_{max} (film)/cm⁻¹ 1570 (C=C); $\delta_{\rm H}$ (CDCl₃) 7.50–7.09 (15 H, m, PhS), 6.25(1 H, q, J6.6, CH=CSPh), 3.82(1 H, d, J1.6, CHSPh), 3.26-3.17 (1 H, q, J 5.9, CHMe), 1.79 (3 H, d, J 6.6, MeCH=CSPh), 1.36 (3 H, d, J 5.9, CHMe) and 1.24 (6 H, s, PhSCMe₂); $\delta_{\rm C}({\rm CDCl}_3)$ 16.0 (CHMe), 16.3 (MeCH=CSPh), 27.9, 29.4 (PhSCMe₂), 41.0 (CHMe), 55.5 (CSPh), 64.7 (CHSPh), 125.6, 125.9, 128.3, 128.4, 128.9, 130.5, 134.2, 137.7 (ArC, MeC-H=CSPh), 132.3, 136.2, 137.1 and 138.6 (ArC-S, MeCH=CSPh): m/z 450 (1%, M⁺), 341 (100, M⁺ – PhS), 231 (28, M⁺ PhSH – PhS), 218 (13, PhSSPh), 189 (28, C₁₂H₁₃S), 163 (20, Me₂C=CPhS), 151 (45, Me₂CSPh), 149 (32, MeCH=CSPh) and $121(55, M^+ - PhSH - PhS - PhSH)$ (Found: M⁺, 450.1489, $C_{27}H_{30}S_3$ requires *M*, 450.1503).

Rearrangement by Method B: Typical Procedure.—Thionyl chloride (0.090 cm³) was added to an ice cooled soluion of the syn-alcohol 12b (432 mg, 0.73 mmol) in carbon tetrachloride (9 cm^3) and triethylamine (0.6 cm^3) kept in a foil-wrapped flask. After 1 min, the mixture was poured into dilute hydrochloric acid (15 cm³), extracted with dichloromethane $(3 \times 5 \text{ cm}^3)$ and the combined extracts were washed with water $(3 \times 10 \text{ cm}^3)$, dried (MgSO₄) and evaporated to dryness. The crude product was submitted to flash column chromatography on silica gel, eluting with hexane-dichloromethane (3:1), to give three products: a 1.3:1 mixture of the vinyl sulfides (E)and (Z)-(4RS,5RS)-4,6-dimethyl-3,5,6-tris(phenylthio)hept-2ene 17b (170 mg, 52%), as an oil, $R_{\rm F}$ [hexane-CH₂Cl₂ (3:1)] 0.46; $v_{max}(film)/cm^{-1}$ 1570 (C=C); $\delta_{H}(CDCl_{3})$ 7.60–7.03 (15 H, m, PhS), 6.25 (Z), 5.49 (E) (1H, 2q, J 6.6 and 7, CH=CSPh), 3.93 (E), 3.23 (Z) [1 H, dq, J 2.5 and 7.0, and q, J 5.9, CHMe), 3.82 (Z), 3.75 (E) (1 H, 2d, J 1.6 and 2.5, CHSPh), 1.89 (E), 1.79 (Z) (3 H, 2 d, J7 and 6.6, MeCH=CSPh), 1.38 (E), 1.23 (Z), 1.17 (E) (6 H, 3 s, PhSCMe₂) and 1.36 (3 H, d, J 7, CHMe); $\delta_{\rm C}({\rm CDCl}_3)$ 14.8 (E), 16.0 (Z) (CHMe), 16.3 (Z), 18.3 (E) (MeCH=CSPh), 25.6 (E), 27.9 (Z), 29.3 (E), 29.5 (Z) (PhSCMe₂), 37.0 (E), 41.0 (Z) (CHMe), 55.5 (Z), 56.9 (E) (CSPh), 61.6 (E), 64.7 (Z) (CHSPh), 125.6, 125.9, 126.8, 127.1, 128.3, 128.5, 128.6, 128.9, 129.2, 130.5, 131.9, 134.2, 137.7, 137.8 (ArC, MeC=CH), 132.3, 135.7, 136.2, 137.1, 138.6 and 141.5 (ArC-S, CH=CSPh); m/z 450 (0.1%, M⁺), 341 (87, M⁺ – PhS), 232 (100, M^+ – PhS – PhS), 231 (67, M^+ – PhSH – PhS), 218 (70, PhSSPh), 186 (65, C₁₉H₁₁S), 163 (20, Me₂C=CPhS), 149 (22, MeCH=CSPh) and 121 (22, M⁺ PhSH – PhS – PhSH) (Found: M^+ , 450.1494. $C_{27}H_{30}S_3$ requires M, 450.1503); and the alkene 2,4-dimethyl-5,5bis(phenylthio)hept-2-ene 18b (62 mg, 25%) as an oil $R_{\rm F}$ [hexane-dichloromethane (3:1)] 0.58; $v_{max}(film)/cm^{-1}$ 1570 (C=C); δ_H(CDCl₃) 7.60–7.21 (10 H, m, PhS), 5.39 (1 H, qd, J 1.3 and 9.7, CH=CMe₂), 2.89 (1 H, dq, J 9.7 and 6.7, CHMe), 1.67, 1.55 (6 H, 2 s, CH=CMe₂), 1.44, 1.43 (2 H, 2 q, J 7 and 7.3, CH₂Me), 1.15 (3 H, d, J 6.7, CHMe) and 0.98 (3 H, t, J 7.1, CH_2Me); $\delta_C(CDCl_3)$ 9.0 (CH₂Me), 17.2 (CHMe), 18.0, 26.0 $(CH=CMe_2)$, 30.1 (CH_2Me) , 40.5 (CHMe), 75.2 $[C(SPh)_2]$, 126.3, 127.5, 128.3, 128.3, 128.6, 128.7, 129.0, 137.1, 137.2 (ArC, CH=CMe₂), 131.8 (CH=CMe₂), 132.7 and 132.8 (ArC-S); $m/z 259 (1\%, M^+ - C_6 H_{11}), 232 (33, M^+ - PhS), 110 (100, M^+)$ PhSH) and 109 (98, PhS) (Found: $M^+ - C_6 H_{11}$, 259.0623. $C_{15}H_{15}S_2$ requires $M - C_6H_{11}$, 259.0612).

Rearrangement of (3RS,4RS)–2,4,6-*Trimethyl*-3,5,5-*tris*-(*phenylthio*)*heptan*-2-*ol* **16c**.—By method A, the *syn*-alcohol **16c**

(42 mg, 0.087 mmol) and TsOH (4 mg, 0.023 mmol) in dichloromethane (0.4 cm³) gave the *vinyl sulfide* (4RS,5RS)-2,4,6-trimethyl-3,5,6-tris(phenylthio)hept-2-ene **17c** (24 mg, 60%), as an oil $R_{\rm F}$ [hexane-dichloromethane (3:1)] 0.41; $v_{\rm max}$ (film)/cm⁻¹ 1550 (C=C); $\delta_{\rm H}$ (CDCl₃) 7.49–7.00 (15 H, m, PhS), 3.96 (1 H, dq, J 3.4 and 6.9, CHMe) 3.61 (1 H, d, J 3.4, CHSPh), 2.08, 1.90 (6 H, 2 s, C=CCMe₂), 1.38, 1.23 (6 H, 2 s, PhSCMe₂) and 1.29 (3 H, d, J 6.9, CHMe); m/z 464 (1%, M⁺), 235 (100, M⁺ – PhS), 245 (35, M⁺ – PhSH – PhS), 218 (16, PhSH), 203 (48, M⁺ – PhSH – PhS – C₃H₆), 191 (31, C₁₂H₁₅S), 164 (29, Me₂C=CHPhS), 163 (54, Me₂C=CPhS), 151 (35, Me₂CSPh), 149 (36, MeCH=CSPh), 135 (36, M⁺ – PhSH – PhS – PhSH – PhS – PhSH), 121 (45, PhSC), 110 (82, PhSH) and 109 (48, PhS) (Found: M⁺, 464.1642. C₂₈H₃₂S₃ requires M, 464.1659).

By method B, the *syn*-alcohol **16c** (64 mg, 0.013 mmol), triethylamine (0.11 cm³) and thionyl chloride (0.016 cm³) in carbon tetrachloride (1.8 cm³) gave, as the only product, the alkene **17c** (50 mg, 82%).

Rearrangement of (3RS,4RS)-2,4-Dimethyl-3,5,5-tris(phenylthio)octan-2-ol 16d.—By method A, the syn-alcohol 16d (33 mg, 0.068 mmol) and TsOH (3 mg, 0.0017 mmol) in dichloromethane (0.3 cm³) gave the vinyl sulfide (Z)-(5RS,6RS)-5,7dimethyl-4,6,7-tris(phenylthio)oct-2-ene 17d (20 mg, 64%), as an oil, $R_{\rm F}$ (carbon tetrachloride) 0.60; $v_{\rm max}({\rm film})/{\rm cm}^{-1}$ 1570 (C=C); δ_H(CDCl₃) 7.63–7.10 (15 H, m, PhS), 6.16 (1 H, t, J 7, CH=CSPh), 3.83 (1 H, s, CHSPh), 3.20 (1 H, a, J7, CHMe), 2.37 (1 H, dq, J7 and 7.5, CH_AH_BCH=CSPh), 2.16 (1 H, dq, J7 and 7.5, CH_AH_BCH=CSPh), 1.36 (3 H, d, J 7, CHMe), 1.23, 1.22 (6 H, 2 s, PhSCMe₂) and 0.99 (3 H, t, J 7.5, CH₂Me); m/z 355 $(18\%, M^+ - PhS), 245 (50, M^+ - PhS - PhSH), 163 (27,$ Me₂C=CPhS), 151 (42, Me₂CSPh) 135 (27, M⁺ - PhSH -PhS - PhSH), 110 (100, PhSH) and 109 (65, PhS) (Found: $M^+ - C_6H_5S$, 355.1578. $C_{22}H_{27}S_2$ requires $M - C_6H_5S$, 355.1548).

By method B, the syn-alcohol 16d (35 mg, 0.073 mmol), triethylamine (0.060 cm³) and thionyl chloride (0.009 cm³) in carbon tetrachloride (1 cm³) gave three products: a 2.5:1 mixture of the vinyl sulfides (E)- and (Z)-(5RS,6RS)-5,7dimethyl-4,6,7-tris(phenylthio)oct-2-ene 17d (23 mg, 68%), as an oil, $R_{\rm F}$ (carbon tetrachloride) 0.60; $v_{\rm max}({\rm film})/{\rm cm}^{-1}$ 1570 (C=C); $\delta_{\rm H}$ (CDCl₃) 7.50–7.03 (15 H, m, PhS), 6.17 (Z), 5.40 (E) (1 H, 2 t, J7 and 7.3, CH=CSPh), 3.92 (E), 3.19 (Z) (1 H, dq, J2 and 7, and q, J 7, CHMe), 3.83 (Z), 3.75 (E), (1 H, s, d, J 2, CHSPh), 2.38 (Z), 2.15 (Z) (2H, 2 dq, J 7.3, 7.4 and J 7, 7.6, CH₂CHCSPh), 1.37 (E), 1.36 (Z), (3 H, 2 d, J7, CHMe), 1.23 (Z), 1.22 (Z), 1.17 (E) (6 H, 3 s, PhSCMe₂), 0.98 (E) and 0.91 (Z) (3 H, 2 t, J 7.4 and 7.6, CH_2Me); m/z 355 (100%, M⁺ PhS), 245 (50, M⁺ – PhS – PhSH), 163 (27, Me₂C=CPhS), 151 (45, Me₂CSPh), 110 (82, PhSH) and 109 (39, PhS) (Found: $M^+ - C_6H_5S$, 355.1560. $C_{22}H_{27}S_2$ requires $M - C_6H_5S$, 355.1548).

Rearrangement of (3RS,4RS)-2,4-Dimethyl-3,5,5-tris(phenylthio)oct-7-en-2-ol **16e**.—By method A, the syn-alcohol **16e** (171 mg, 0.35 mmol) and TsOH (15 mg) in dichloromethane (1.5 cm³) gave a complex mixture of products. However, by method B, the syn-alcohol **16e** (131 mg, 0.25 mmol), triethylamine (0.20 cm³) and thionyl chloride (0.035 cm³) in carbon tetrachloride (3.5 cm³) gave as the only identifiable product the alkene 5,7dimethyl-4,4-bisphenylthioocta-1,6-diene **18e** (24 mg, 25%), as an oil, R_F [hexane–dichloromethane (3:1)] 0.53; v_{max} (film)/cm⁻¹ 1570 (C=C); δ_H (CDCl₃) 7.66–7.18 (10 H, m, PhS), 6.11 (1 H, ddt, J 17.1, 10.3 and 6.6, CH₂CH=CH₂), 5.39 (1 H, q, J 9.7, CH=CMe₂), 5.06 (1 H, dd, J 1.8 and 10.3, CH₂CH=CH_AH_B), 4.93 (1 H, dd, J 1.8 and 17, CH₂CH=CH_AH_B), 2.88 (1 H, dq, J 9.7 and 6.8, CHMe), 2.24 (2 H, d, J 6.6, CH₂CH=CH₂), 19.68, 1.50 (6 H, 2 s, CH=C Me_2) and 1.16 (3 H, d, J 6.7, CHMe); $\delta_{\rm C}$ (CDCl₃) 17.2 (CHMe), 18.2, 26.0 (CH=C Me_2), 41.3 (CHMe), 41.6 (CH₂CH=CH₂), 73.0 [C(SPh)₂], 117.4 (CH=CH₂), 126.1, 127.1, 127.4, 128.4, 128.8, 128.9, 129.0, 134.0, 137.1, 137.2 (ArC, CH=CMe₂, CH₂CH=CH₂), 132.2 (CH=CMe₂), 132.4 and 132.5 (ArC-S); m/z 271 (4%, M⁺ – C₆H₁₁), 245 (48, M⁺ – PhS), 228 (30, C₁₅H₁₆S), 134 (28, C₁₀H₁₄), 119 (35, C₉H₁₁), 110 (100, PhSH) and 109 (52, PhS) (Found: M⁺ – C₆H₁₁, 271.0623. C₁₆H₁₅S₂ requires $M - C_6H_{11}$, 271.0612).

Rearrangement of (3RS,4RS)-2,4-Dimethyl-3,5,5-tris(phenylthio)nonan-2-ol 16f.—By method A, the syn-alcohol 16f (192 mg, 0.38 mmol) and TsOH (17 mg, 0.1 mmol) in dichloromethane (1.7 cm^3) gave the vinyl sulfide (Z)-(6RS,7RS)-6,8-dimethyl-5,7,8-tris(phenylthio)non-2-ene 17f (90 mg, 49%), as oil $R_{\rm F}$ [hexane–CH₂Cl₂ (3:1)] 0.50; ν_{max} (film)/cm⁻¹ 1570 (C=C); $\delta_{\rm H}$ (CDCl₃) 7.48–7.07 (15 H, m, PhS), 6.19 (1 H, t, J 6.6, CH=CSPh), 3.84 (1 H, d, J 1.5, CHSPh), 3.22 (1 H, q, J 6.9, CHMe), 2.35 (1 H, dq, J 7.6 and 15, CH_AH_BCH=CSPh), 2.13 (1 H, dq, J 6.6 and 15, CH_AH_BCH=CSPh), 1.57-1.25 (2 H, m, CH₂Me), 1.36 (3 H, d, J 6.9, CHMe), 1.22, 1.21 (6 H, 2 s, PhSCMe₂) and 0.86 (3 H, t, J 7.4, CH₂Me); δ_{c} (CDCl₃) 14.0 (CH₂Me), 16.5 (CHMe), 22.5 (CH₂Me), 27.8, 29.5 (PhSCMe₂), 32.4 (CH₂CH₂Me), 40.6 (CHMe), 51.8 (CSPh), 64.4 (CHSPh), 125.6, 125.8, 128.4, 128.6, 128.6, 128.9, 130.1, 137.7 (ArC, MeCH=CSPh), 132.4, 136.3, 136.3, 138.7 and 139.7 (ArC-S, MeCH=CSPh); m/z 478 (0.3%, M⁺), 369 (61, M⁺ – PhS), 259 (59, $M^+ - PhS - PhSH$), 217 (22, $C_{14}H_{17}S$), 205 (33, C₁₃H₁₇S), 163 (72, Me₂C=CPhS), 151 (100, Me₂CSPh), 123 (35, PhSCH₂), 121 (28, PhSC), 110 (65, PhSH), 109 (54, PhS) and 107 (70, C_8H_{18}) (Found: M⁺, 478.1818. $C_{29}H_{34}S_3$ requires M, 478.1815).

By method B, the syn-alcohol 16f (225 mg, 0.45 mmol), triethylamine (0.37 cm^3) and thionyl chloride (0.056 cm^3) in carbon tetrachloride (5.6 cm³) gave a 1.8:1 mixture of the vinyl sulfides (E)- and (Z)-(6RS,7RS)-6,8-dimethyl-5,7,8-tris(phenylthio)non-2-ene 17f (105 mg, 49%), as an oil, R_F [hexanedichloromethane (3:1)] 0.50; $v_{max}(film)/cm^{-1}$ 1570 (C=C); $\delta_{\rm H}({\rm CDCl}_3)$ 7.60–7.04 (15 H, m, PhS), 6.18 (Z), 5.45 (E) (1 H, 2 t, J 6.6 and 7.3, CH=CSPh), 3.93 (E), 3.22 (Z) (1 H, dq, J 2.4 and 7.3, and q, J7, CHMe), 3.84(Z), 3.75(E) (1H, 2d, J1.5 and 2.4, CHSPh), 2.38-2.14 (2 H, m, CH₂CH=CSPh), 1.47-1.12 (2 H, m, CH_2Me , 1.38 (E), 1.36 (Z) (3 H, 2 d, J7, CHMe), 1.22 (Z), 1.21 (Z), 1.18 (E) (6 H, 3 s, PhSCMe₂) and 0.89 (3 H, t, J 7.4, CH_2Me); $\delta_C(CDCl_3)$ 13.9(Z), 15.3(E)(CH_2Me), 16.5(Z), 18.6 (E) (CHMe), 22.5 (Z), 22.7 (E) (CH₂Me), 25.6 (E), 27.7 (Z), 29.5(E), 29.5(Z) (PhSCMe₂), 31.1(E), 32.4(Z) (CH₂CH₂Me), 37.3 (E), 40.6 (Z) (CHMe), 55.7 (Z), 57.0 (E) (CSPh), 61.7 (E), 64.4(Z)(CHSPh), 125.5, 125.6, 125.8, 126.7, 128.0, 128.3, 128.3, 128.5, 128.6, 128.7, 128.9, 129.2, 130.1, 130.2, 131.6, 133.8, 136.3, 136.9, 137.1, 137.6, 137.7, 139.6 (ArC, MeCH=CSPh), 131.9, 132.3, 136.1, 136.2, 138.6, 138.7 and 140.5 (ArC-S, MeCH= CSPh); m/z 478 (1%, M⁺), 369 (100, M⁺ – PhS), 327 (13, $M^+ - Me_2CSPh$), 259 (35, $M^+ - PhS - PhSH$), 205 (28, C₁₃H₁₇S), 163 (50, Me₂C=CPhS), 151 (76, PhSCMe₂), 123 (23, PhSCH₂) and 109 (39, PhS) (Found: M^+ , 478.1785. $C_{29}H_{34}S_3$ requires *M*, 478.1815); and the *alkene* 2,4-dimethyl-5,5-bis-(phenylthio)non-2-ene 18f (45 mg, 27%), as an oil R_F [hexanedichloromethane (3:1)] 0.60; $v_{max}(film)/cm^{-1}$ 1570 (C=C); $\delta_{\rm H}({\rm CDCl}_3)$ 7.65–7.18 (10 H, m, PhS), 5.38 (1 H, qd, J 1.2 and 9.7, CH=CMe₂), 2.89 (1 H, dq, J 9.7 and 6.8, CHMe), 1.67 and $1.54 (6 \text{ H}, 2 \text{ s}, \text{CH}=\text{CM}e_2), 1.50-1.36 [4 \text{ H}, \text{m}, (\text{CH}_2)_2\text{CH}_2\text{M}e],$ 1.13 (3 H, d, J 6.8, CHMe), 1.07-1.02 (2 H, m, CH₂Me) and 0.78 (3 H, t, J 7.1, CH₂Me); δ_C(CDCl₃) 14.0 (CH₂Me), 17.3 (CHMe), 18.1, 26.0 $(CH=CMe_2)$, 22.7 (CH_2Me) , 26.3 (CH₂CH₂Me), 37.2 [CH₂(CH₂)₂Me], 41.2 (CHMe), 74.76 [C(SPh)₂], 126.4, 127.2, 127.5, 128.3, 128.7, 128.8, 129.1, 137.0,

137.1 (ArC,CH=CMe₂), 131.9 (CH=CMe₂), 132.8 and 133.0 (ArC-S); m/z 369 (3.5%, M⁺), 287 (10, C₆H₁₀), 261 (100, M⁺ – PhS), 151 (35, M⁺ – PhS – PhS), 135 (16, M⁺ – PhS – PhSH – Me), 109 (53, PhS) and 95 (67, C₇H₁₁) (Found: M⁺, 369.1699. C₂₃H₂₉S₂ requires *M*, 369.1704).

Rearrangement of (3RS,4RS)-2,4-Dimethyl-7-phenyl-3,5,5tris(phenylthio)heptan-2-ol 16g.—By method A, the syn-alcohol 16g (366 mg, 0.67 mmol) and TsOH (30 mg, 0.17 mmol) in dichloromethane (1.7 cm³) gave the vinyl sulfide (Z)-(4RS,-5RS)-4,6-dimethyl-1-phenyl-3,5,6-tris(phenylthio)hept-2-ene 17g (166 mg, 47%), as oil $R_{\rm F}$ [hexane-dichloromethane (3:1)] $0.41; \nu_{max}(film)/cm^{-1} 1575 (C=C); \delta_{H}(CDCl_{3}) 7.65-7.11 (20 H, m, J)$ PhS, Ph), 6.34 (1 H, t, J 6.7, CH=CSPh), 3.88 (1 H, d, J 1.4, CHSPh), 3.81 (1 H, dd, J8 and 15.7, CH_AH_BCH=CSPh), 3.43 (1 H, dd, J 6.2 and 15.7, $CH_AH_BCH=CSPh$), 3.29 (1 H, q, J 6.9, CHMe), 1.36 (3 H, d, J 6.9, CHMe), 1.23 and 1.21 (6 H, 2 s, PhSCMe₂); $\delta_{C}(CDCl_{3})$ 16.5 (CHMe), 27.6, 29.5 (PhSCMe₂), 36.6 (PhCH₂), 40.7 (CHMe), 55.8 (CSPh), 64.0 (CHSPh), 125.8, 125.9, 126.0, 127.2, 128.4, 128.5, 128.7, 128.9, 129.0, 130.3, 132.4, 137.7 (ArC, MeCH=CSPh), 130.0, 135.9, 137.5, 138.6 and 140.3 (ArC-S and MeCH=CSPh); m/z 526 (0.1%, M⁺), 417 (47, M^+ – PhS), 375 (18, M^+ – Me₂CSPh), 307 (30, M^+ – PhS – PhSH), 265 (13, $C_{18}H_{17}S$), 217 (19, $C_{14}H_{17}S$, 163 (34, Me₂C=CPhS), 151 (76, Me₂CSPh), 143 (45, C₁₁H₁₁), 109 (26, PhS) and 91 (100, PhCH₂) (Found: M⁺, 526.1832. C₃₃H₃₄S₃ requires M, 526.1815).

By method B, the syn-alcohol 16g (100 mg, 0.18 mmol), triethylamine (0.15 cm³) and thionyl chloride (0.023 cm³) in carbon tetrachloride (2.3 ml) gave a 2.2:1 mixture of the vinyl sulfides (E)- and (Z)-(4RS,5RS)-4,6-dimethyl-1-phenyl-3,5,6tris(phenylthio)hept-2-ene 17g (66 mg, 52%), as an oil, $R_{\rm F}$ [hexane-dichloromethane (3:1)] 0.41; $v_{max}(film)/cm^{-1}$ 1575 (C=C); δ_H(CDCl₃) 7.56–7.10 (20 H, m, PhS and Ph), 6.38 (Z), 5.59 (E) (1 H, 2t, J 6.8 and 7.5, CH=CSPh), 4.10 (E), 3.32 (Z) (1 H, dq, J 2.2 and 7, and q, J 6.8, CHMe), 3.92 (Z), 3.85 (E) (1 H, 2 d, J1.4 and 2.2, CHSPh), 3.90-3.79 (1 H, m, CH_AH_BCH=CSPh), 3.69 (E), 3.43 (Z) (1 H, 2 dd, J 7.2 and 16.3 and J 6.2 and 16.3, CH_AH_BCH=CSPh), 1.43 (E), 1.36 (Z) (3 H, 2 d, J 7.1 and 6.9, CHMe), 1.26 (E), 1.24 (E) and 1.23 (Z) (6 H, 3 s, PhSCMe₂); δ_c(CDCl₃) 16.5 (Z), 18.5 (E) (CHMe), 25.4 (E), 27.6 (Z), 29.4 (E), 29.5 (Z) (PhSCMe₂), 35.0 (E), 36.6 (Z) (CH₂Ph), 37.3 (E), 40.7 (Z) (CHMe), 55.7 (Z), 57.1 (E) (CSPh), 61.6 (E), 64.0 (Z) (CHSPh), 125.6, 125.8, 125.9, 127.2, 128.4, 128.5, 128.6, 128.9, 129.0, 129.2, 129.5, 130.0, 132.5, 136.8, 137.7, 137.8 (ArC, MeCH=CSPh), 131.8, 132.3, 135.1, 135.8, 137.5, 138.5, 138.6, 140.2, 140.4 and 142.6 (ArC-S, MeCH=CSPh); m/z 526 (0.3%, M^+), 417 (70, M^+ – PhS), 375 (11, M^+ – Me₂CSPh), 309 $(39, M^+ + 1 - PhS - PhS), 252 (13, C_{17}H_{16}S), 151 (17,$ Me₂CSPh), 143 (39, C₁₁H₁₁), 109 (26, PhS) and 91 (100, PhCH₂) (Found: M^+ , 526.1840. $C_{33}H_{34}S_3$ requires *M*, 526.1815) and the alkene 2,4-dimethyl-7-phenyl-5,5-bis(phenylthio)non-2-ene 18g (145 mg, 34%), as an oil R_F [hexane-CH₂Cl₂ (3:1)] 0.53; $v_{max}(film)/cm^{-1}$ 1575 (C=C); $\delta_{H}(CDCl_{3})$ 7.70-6.83 (15 H, m, PhS), 5.41 (1 H, qd, J 1.3 and 9.6, CH=CMe₂), 3.00 (1 H, dq, J9.6 and 6.7, CHMe), 2.85-2.78 (2 H, m, CH₂CH₂Ph), 1.88-1.81 (2 H, m, CH₂CH₂Ph), 1.69, 1.58 $(6 \text{ H}, 2 \text{ s}, \text{CH}=\text{C}Me_2)$ and $1.23 (3 \text{ H}, d, J 6.8, \text{CH}Me); \delta_{\text{C}}(\text{CDCl}_3)$ 17.4 (CHMe), 18.0, 25.9 (CH=CMe₂), 30.6 (CH₂CH₂Ph), 39.3 (CH₂CH₂Ph), 41.5 (CHMe), 73.8 [C(SPh)₂], 125.7, 126.1, 128.2, 128.4, 128.8, 136.7 (ArC, CH=CMe₂), 132.1 (CH=CMe₂), 132.6, 132.7 and 141.7 (ArC-S, ArC-C); m/z 418 (0.5%, M⁺), 417 $(1.3, M^+ - 1), 335 (1, M^+ - C_6 H_{11}), 309 (43, M^+ - PhS), 251 (22, C_{17}H_{16}S)$ 143 (43, $C_{11}H_{11}$), 109 (19, PhS) and 91 (100, PhCH₂) (Found: M⁺, 418.1772. $C_{27}H_{30}S_2$ requires M, 418.1772).

A portion of this 2.2:1 mixture of the (E) and (Z) vinyl sulfides 17g (55 mg, 0.10 mmol) was dissolved in dichloro-

methane (0.5 cm³), and TsOH (4.4 mg) was added to the solution. The mixture was then refluxed for 15 min after which work-up gave the vinyl sulfide (Z)-(4RS,5RS)-4,6-dimethyl-1-phenyl-3,5,6-tris(phenylthio)hept-2-ene 17g (39 mg, 71%) as the only reaction product.

Rearrangement of 2-Methyl-7-phenyl-3,5,5-tris(phenylthio)heptan-2-ol 16h.—By method A, the syn-alcohol 16g (134 mg, 0.25 mmol) and TsOH (12 mg, 0.069 mmol) in dichloromethane (1.2 cm^3) gave a 2.8:1 mixture of the vinyl sulfides (Z)and (E)-6-methyl-1-phenyl-3,5,6-tris(phenylthio)hept-2-ene 17h (61 mg, 47%), as an oil, R_F [hexane-dichloromethane (2:1)] 0.35; $\nu_{max}(\text{film})/\text{cm}^{-1}$ 1575 (C=C); $\delta_{H}(\text{CDCl}_{3})$ 7.46–7.07 (20 H, m, PhS, Ph), 6.31 (Z), 6.19 (E), (1 H, 2 t, J 7.3 and 7, CH=CSPh), 3.90–3.62 (3 H, m, CHSPh, CSPhCH₂CSPh), 3.40 (Z), 3.14 (E) (1 H, 2 d, J 14.5 and 14.3, CH_AH_BCH=CSPh), 2.96 (E), 2.47 (Z) (1 H, 2 dd, J 11.5 and 14.3 and J 11.5 and 14.5, CH_AH_BCH=CSPh), 1.22 and 1.21 (6 H, 2 s, PhSCMe₂); $\delta_{\rm C}({\rm CDCl}_3)$ 25.2 (Z), 25.4 (E), 29.1 (Z), 29.2 (E) (PhSCMe_2), 33.2 (E), 40.3 (Z) (CSPhCH₂CSPh), 35.8 (E), 36.5 (Z) (PhCH₂CH=CSPh), 54.7 (Z), 54.9 (E) (PhSCMe₂), 58.7 (E), 59.9 (Z) (CHSPh), 128.7, 128.8, 129.0, 130.1, 130.3, 130.6, 137.6, 138.5, 139.9, 140.1 (ArC, PhCH₂CH=CSPh), 131.0, 131.9, 132.4, 135.1, 135.2, 138.1 and 139.9 (ArC-S, ArC-C, PhCH₂CH=CSPh); m/z 403 (1.5%, M⁺ – PhSH), 293 (85, $M^{+} - PhSH - PhSH$), 203 (62, $C_7H_{10}SPh$), 110 (100, PhSH), 109 (50, PhS) and 91 (95, PhCH₂) (Found: $M^+ - C_6H_6S$, 403.1576. $M - C_6 H_6 S$ requires 403.1548); and a 1.7:1 mixture of the vinyl sulfides (Z)- and (E)-6-methyl-1-phenyl-3,5,6tris(phenylthio)hept-3-ene 19h (16 mg, 12%), as an oil, $R_{\rm F}$ [hexane-CH₂Cl₂ (2:1)] 0.32; $\nu_{max}(film)/cm^{-1}$ 1575 (C=C); $\delta_{\rm H}({\rm CDCl}_3)$ 7.66:6.75 (20 H, m, PhS, Ph), 5.87 (Z), 5.85 (E) (1 H, 2 d, J 10.3 and 11.2, CH=CSPh), 4.57 (Z), 3.98 (E) (1 H, 2 d, J 10.3 and 11.2, CHSPh), 2.61 (Z), 2.46 (E), 2.44 (Z) (2 H, 3 t, J7.5, 12.1, 12.1, CH₂CH₂Ph), 2.30 (Z), 2.10 (E), 1.97 (Z) (2 H, t, and 2 dt, J7.5, J 5 and 12.1, J 5 and 12.1, CH₂CH₂Ph), 1.30 and 1.26 (6 H, 2 s, PhSCMe2); m/z 512 (0.1%, M+), 403 (48, M+ - PhSH), 293 (41, M^+ - PhSH - PhSH), 203 (62, C₇H₁₀SPh) and 91 (100, PhCH₂) (Found: M^+ , 512.1717. C₃₂H₃₂S₃ requires *M*, 512.1659).

By method B, the syn-alcohol **16h** (190 mg, 0.36 mmol), triethylamine (0.295 cm³) and thionyl chloride (0.045 cm³) in carbon tetrachloride (4.5 cm³) gave a 2.7:1 mixture of the vinyl sulfides (Z)- and (E)-6-methyl-1-phenyl-3,5,6-tris(phenylthio)-hept-2-ene **17h** (97 mg, 53%) and a 1.8:1 mixture of the vinyl sulfides (Z)- and (E)-6-methyl-1-phenyl-3,5,6-tris(phenylthio)-hept-3-ene **19h** (13 mg, 8%).

Methyl (2RS,3RS)-2,3,5-Trimethyl-4,4-bis(phenylthio)-

hexanoate 24c.—Butyllithium (1.6 mol dm⁻³ solution in hexane; 4.3 cm³, 6.9 mmol) was added to a stirred solution of 2-methyl-1,1-bis(phenylthio)propane 9c (1.58 g, 5.76 mmol) in dry THF (60 cm³) and TMEDA (0.9 cm³, 6.45 mmol) under argon at 0 °C. After 15 min, the solution was cooled to -78 °C and methyl crotonate (0.52 g, 5.2 mmol) in dry THF (5 cm³) was added dropwise to the mixture which was then stirred at this temperature for 1 h. Subsequent methylation of the enolate was accomplished by addition of methyl iodide (2.6 mol dm^{-3} solution in THF; 6 cm³, 16.5 mmol) whilst the mixture was stirred for 15 min at -78 °C and then 2 h at room temperature. The reaction was then quenched by addition of saturated aqueous ammonium chloride (20 cm³) to the mixture which was then extracted with ether $(3 \times 20 \text{ cm}^3)$. The combined extracts were dried (MgSO₄) and evaporated to dryness to give an oil that was purified by flash column chromatography on silica gel, eluting with 1:1 hexane-dichloromethane, to give the ester syn-24c (1.1 g, 51%) as an oil, R_F [hexane-dichloromethane (1:1)] $0.45; v_{max}(CDCl_3)/cm^{-1}$ 1720 (CO) and 1580 (PhS); $\delta_{\rm H}(CDCl_3)$:

7.74–7.23 (10 H, m, PhS), 3.62 (3 H, s, CO₂Me), 3.50 (1 H, dq, *J* 3 and 7.3, C*H*CO₂Me), 2.91 (1 H, dq, *J* 3 and 7.1, C*H*Me), 2.41 (1 H, 2 q, *J* 6.6, C*H*Me₂), 1.37 (3 H, d, *J* 7.3, *Me*CHCO₂Me), 1.14, 1.10 (6 H, 2 d, *J* 6.6, CH*Me*₂) and 1.05 (3 H, d, *J* 7.1, CH*Me*); $\delta_{\rm C}$ (CDCl₃) 12.9 (*Me*CHCO₂Me), 14.5 (CH*Me*), 19.4, 20.30 (CH*Me*₂), 37.3 (CHMe₂), 40.7 (MeCHCO₂Me), 41.92 (CHMe), 51.7 (CO₂*Me*), 79.1 [*C*(SPh)₂], 128.3, 128.5, 128.7, 128.9, 129.0, 135.9, 137.1 (ArC), 132.8, 133.8 (ArC-S) and 177.5 (CO₂Me); *m*/*z* 278 (11%, M⁺ – PhSH), 247 (15, M⁺ – PhS – OMe), 169 (55, M⁺ – PhSH – PhS), 110 (100, PhSH) and 109 (53, PhS) (Found: M⁺ – C₆H₆S, 278.1328. C₁₆H₂₂O₂S requires *M* – C₆H₆S, 278.1335).

Methyl (2RS,3RS)-2,3-Dimethyl-4,4-bis(phenylthio)-

heptanoate 24d.—In the same way, 1,1-bis-phenylthiobutane 9d (150 mg, 0.64 mmol), TMEDA (0.085 cm³, 0.61 mmol), butyllithium (1.6 mol dm⁻³ solution in hexane; 0.48 cm³, 0.77 mmol) in dry THF (6 cm³), methyl crotonate (49 mg, 0.49 mmol) in dry THF (0.5 cm³) and methyl iodide (2.6 mol dm⁻³ solution in THF; 0.6 cm³, 1.56 mmol) gave the ester syn-24d (95 mg, 44%) as an oil, $R_{\rm F}$ [hexane-dichloromethane (1:1)] 0.28; $v_{\rm max}/{\rm cm^{-1}}$ 1720 (CO) and 1580 (PhS); $\delta_{\rm H}({\rm CDCl}_3)$ 7.60–7.19 (10 H, m, PhS), 3.66 (3 H, s, CO₂Me), 3.32 (1 H, dq, J2.1 and 7.2, CHCO₂Me), 2.70 (1 H, dq, J 2.1 and 7.1, CHMe), 1.55-1.45 (2 H, m, CH₂CH₂Me), 1.41 (3 H, d, J7.2, MeCHCO₂Me), 1.39-1.33 (2 H, m, CH₂Me), 1.11 (3 H, d, J7.1, CHMe) and 0.69 (3 H, t, J 6.7, CH_2Me); m/z 279 (21%, M⁺ – PhS), 247 (35, M⁺ – PhS – OMe), 219 (47, M^+ – PhSH – $C_2H_4O_2$), 177 (27, M^+ – PhSH – $C_2H_4O_2 - C_4H_9$), 169 (100, M^+ – PhS – PhSH), 130 (55, C₇H₁₄O₂), 110 (97, PhSH) and 109 (80, PhS) (Found: $M^+ - C_6H_5S$, 279.1414. $C_{16}H_{23}O_2S$ requires $M - C_6H_5S$) C₆H₅S, 279.1413).

(3RS,4RS)-2,3,4,5-*Tetramethyl*-5,5-*bis*(*phenylthio*)*heptan*-2ol **25**c.—By the method described for **16b**, the *syn*-ester **24c** (155 mg, 0.38 mmol) and methyllithium (1.5 mol dm⁻³ solution in ether; 0.6 cm³, 0.9 mmol) in dry ether (5 cm³) gave the *alcohol syn*-**25c** (110 mg, 71%) as an oil $R_{\rm F}$ [dichloromethane–hexane (3:1)] 0.43; $\nu_{\rm max}$ (film)/cm⁻¹ 3600–3300 (OH) and 1580 (PhS); $\delta_{\rm H}$ (CDCl₃) 7.68–7.14 (10 H, m, PhS), 2.77 [1 H, q, J 7.1, MeCHC(SPh)₂], 2.72 (1 H, br s, OH), 2.64 (1 H, q, J 7, MeCHCOH], 2.45 (1 H, 2 q, J 6.6, CHMe₂) 1.31, 1.06 (6 H, 2 s, HOCMe₂), 1.19, 1.17 (6 H, 2 d, J 6.6, CHMe₂), 1.14 (3 H, d, J 7, MeCHCOH) and 0.92 [3 H, d, J 7.1, MeCHC(SPh)₂]; *m*/*z* 168 (11%, M⁺ – PhSH – PhSH), 153 (15, M⁺ – PhSH – PhSH – Me), 110 (100, PhSH), 109 (23, PhS) and 99 (24, C₆H₁₁O) (Found: M⁺ – C₁₂H₁₂S₂, 168.1515. C₁₁H₂₀O requires $M - C_{12}H_{12}S_2$, 168.1509).

(3RS,4RS)-2,3,4-*Trimethyl*-5,5-*bis*(*phenylthio*)*octan*-2-*ol* **25d**.—In the same way, the *syn*-ester **24d** (97 mg, 0.237 mmol) and methyllithium (1.5 mol dm⁻³ solution in ether; 0.35 cm³, 0.52 mmol) in dry ether (3 cm³) gave the *alcohol syn*-**25d** (83 mg, 86%) as an oil $R_{\rm F}$ [dichloromethane–hexane (3:1)] 0.38; $\nu_{\rm max}$ (film)/cm⁻¹ 3600–3300 (OH) and 1580 (PhS) $\delta_{\rm H}$ (CDCl₃) 7.62–7.14 (10 H, m, PhS), 2.64 [1 H, q, J 7, MeCHC(SPh)₂], 2.41 (1 H, q, J 7, MeCHCOH), 1.41–0.99 [5 H, m, (CH₂)₂Me and OH], 1.30, 1.12 (6 H, 2 s, HOCMe₂), 1.25 (3 H, d, J 7.1, *Me*CHCOH), 1.00 [3 H, d, J 7, *Me*CHC(SPh)₂] and 0.64 (3 H, t, J 7.1, CH₂Me).

Rearrangement of (3RS,4RS)-2,3,4,6-*Tetramethyl*-5,5-*bis*-(*phenylthio*)*heptan*-2-*ol* **25c**.—By method A, the alcohol *syn*-**25c** gave a complex mixture of products. However, by method B, the *syn*-alcohol **25c** (45 mg, 0.11 mmol), triethylamine (0.090 cm³) and thionyl chloride (0.014 cm³) in carbon tetrachloride (1.5 cm³) gave the *alkene* (3RS,4RS)-2,3,4,6-tetramethyl-5,5-bis(phenylthio)hept-1-ene **26c** (36 mg, 84%), as an oil, $R_{\rm F}$ (carbon tetrachloride) 0.65; $\nu_{\rm max}$ (film)/cm⁻¹ 1570 (PhS) and 880 (C=CH₂); $\delta_{\rm H}$ (CDCl₃) 7.78–7.20 (10 H, m, PhS), 4.73 (1 H, s, CH_AH_B=C), 4.70 (1 H, s, CH_AH_B=C), 3.01 (1 H, m, MeCHC=CH₂), 2.52–2.40 [2 H, m, MeCHC(SPh)₂ and CHMe₂], 1.71 (3 H, s, CH₂=CMe), 1.27 (3 H, d, J 7, MeCHC=CH₂), 1.20, 1.07 (6 H, 2 d, J 6.6, CHMe₂) and 1.02 [3 H, d, J 7, MeCHC=CH(SPh)₂]; m/z 260 (4%, M⁺ – PhS), 151 (75, M⁺ – PhS – PhS), 149 (80, M⁺ – PhSH – PhSH), 110 (100, PhSH) and 109 (70, PhS) (Found: M⁺ – C₆H₆S, 260.1598. C₁₇H₂₄S requires $M - C_6H_6S$, 260.1593).

Rearrangement of (3RS,4RS)-2,3,4-Trimethyl-5,5-bis(phenylthio)octan-2-ol **25d**.—By method A, the syn alcohol **25d** gave a complex mixture of products. However, by method B, the synalcohol **25d** (15 mg, 0.038 mmol), triethylamine (0.030 cm³) and thionyl chloride (0.005 cm³) in carbon tetrachloride (0.5 cm³) gave the alkene (3*RS*,4*RS*)-2,3,4-trimethyl-5,5-bis(phenylthio)oct-1-ene **26d** (7 mg, 48%), as an oil, R_F (carbon tetrachloride) 0.66; v_{max} (film)/cm⁻¹ 1570 (PhS) and 880 (C=CH₂); δ_H (CDCl₃) 7.63–7.17 (10 H, m, PhS), 4.72 (2 H, d, J 1.4, CH₂=C), 2.94 (1 H, m, MeCHC=CH₂), 2.25 (1 H, dq, J 2 and 7, MeCHC(SPh)₂], 1.73 (3 H, s, CH₂=CH*Me*) 1.64–1.38 (4 H, m, (CH₂)₂Me), 1.32 (3 H, d, J 7, *Me*CHC=CH₂), 1.03 [3 H, d, J 7, *Me*CHC(SPh)₂] and 0.67 [3 H, t, J 7, CH₂*Me*].

(2RS,3RS)-3-Methyl-2,4,4-tris(phenylthio)hexan-1-ol 27b. Typical Procedure.—To a stirred solution of the syn-ester 12b (1.4 g, 3.63 mmol) in dry THF (36 cm³) kept at 0 °C under argon, lithium aluminium hydride (0.32 g, 8.42 mmol) was added. Stirring was continued at this temperature for 2 h after which the excess of hydride was destroyed by addition of icecold brine. The mixture was filtered through Celite, the filtrate was extracted with ether $(3 \times 15 \text{ cm}^3)$ and the combined extracts were dried (MgSO₄) and evaporated to dryness. Purification by flash column chromatography on silica gel, eluting with dichloromethane-hexane (3:1), gave the alcohol syn-27b (1.21 g, 92%), as an oil, $R_{\rm F}$ (dichloromethane) 0.23; $v_{max}(film)/cm^{-1}$ 3600–3100 (OH) and 1575 (PhS); $\delta_{H}(CDCl_{3})$ 7.71-7.18 (15 H, m, PhS), 4.21 (1 H, t, J 6, CHSPh), 3.60 (2 H, d, J 6, CH₂OH), 2.56 (1 H, q, J 7, CHMe), 2.13 (1 H, br s, OH), 1.78 (1 H, dq, J 12.7 and 7.1, CH_AH_BMe), 1.57 (1 H, dq, J 12.7 and 7.1, CH_AH_BMe), 1.26 (3 H, d, J7, CHMe) and 1.06 (3 H, t, J 7.1, CH_2Me); $\delta_C(CDCl_3)$: 9.6 (CH_2Me), 13.1 (CHMe), 30.5 (CH₂), 41.8 (CHMe), 54.1 (CHSPh), 65.8 (CH₂OH), 75.4 [C(SPh)₂], 125.6, 126.9, 127.5, 127.8, 128.2, 128.4, 128.5, 128.7, 128.0, 129.0, 129.2, 129.4, 131.6, 133.2, 133.9, 136.9, 137.2, 137.4, 137.5 (ArC), 131.8, 132.5 and 135.7 (ArC-S); m/z 330 $(5\%, M^+ - PhSH)$, 221 (13, $M^+ - PhSH - PhS)$, 135 (16, PhSCH=CH₂), 110 (100, PhSH) and 109 (26, PhS) (Found: M⁺ $-C_6H_6S$, 330.1107. $C_{19}H_{22}OS_2$ requires $M - C_6H_6S$, 330.1107).

(2RS,3RS)-3-*Methyl*-2,4,4-*tris*(*phenylthio*)*heptan*-1-*ol* **27d**.— In the same way, the *syn*-ester **12d** (1.4 g, 2.90 mmol) and lithium aluminium hydride (0.28 g, 7.36 mmol) in dry THF (30 cm³) gave the alcohol *syn*-**27d** (1.15 g, 87%), m.p. 61–62 °C (from hexane–light petroleum b.p. 30–40 °C); $R_{\rm F}$ (dichloromethane) 0.24; $\nu_{\rm max}$ (film)/cm⁻¹ 3400 (OH) and 1570 (PhS); $\delta_{\rm H}$ (CDCl₃) 7.69–7.20 (15 H, m, PhS), 4.19 (1 H, dt, *J* 1.1 and 6, CHSPh), 3.54 (2 H, d, *J* 6.1, CH₂OH), 2.51 (1 H, dq, *J* 1.1 and 7.1, CHMe), 1.81–1.66 (3 H, m, CH₂CH₂Me and OH), 1.52–1.37 (2 H, m, CH₂CH₂Me), 1.23 (3 H, d, *J* 7.1, CHMe) and 0.69 [3 H, t, *J* 6.6, (CH₂)₂Me]; $\delta_{\rm C}$ (CDCl₃) 13.2 (CH₂Me), 14.2 (CHMe), 18.1 (CH₂Me), 39.8 (CH₂CH₂Me), 42.3 (CHMe), 54.2 (CHPhS), 75.0 [*C*(PhS)₂], 126.9, 128.4, 128.5, 128.7, 129.1, 131.5, 136.6, 137.0 (ArC), 131.9, 132.6 and 135.8 (ArC-S); *m*/z 344 (2%, M⁺ – PhSH), 235 (15, M⁺ – PhSH – PhS), 153 (27, C₈H₉OS), 135 (30, C_8H_8S), 110 (100, PhSH) and 109 (32, PhS) (Found: $M^+ - C_6H_6OS$, 344.1227. $C_{20}H_{24}OS_2$ requires $M - C_6H_6OS$, 344.1263).

(2RS,3RS)-3-Methyl-1-phenyl-2,4,4-tris(phenylthio)hexan-1ol 27g.—In the same way, the syn-ester 12g (2.25 g, 4.13 mmol) and lithium aluminium hydride (0.2 g, 4.5 mmol) in dry THF (40 cm³) gave the *alcohol syn-27g* (1.7 g, 80%), as an oil, $R_{\rm F}$ (dichloromethane) 0.28; $v_{\rm max}$ (film)/cm⁻¹ 3620–3120 (OH) and 1575 (PhS); $\delta_{\rm H}$ (CDCl₃) 7.79–6.73 (20 H, m, PhS and Ph), 4.26 (1 H, t, J 6, CHSPh), 3.59 (2 H, d, J 6.9, CH₂OH), 2.97 (1 H, dt, J 4 and 13.5, CH₂CH_AH_BPh), 2.79 (1 H, dt, J 5 and 12.7, CH₂CH_AH_BPh), 2.61 (1 H, q, J7, CHMe), 2.19 (1 H, dt, J5 and 14.7, CH_AH_BCH₂Ph), 2.00–1.76 (2 H, m, CH_AH_BCH₂Ph and OH) and 1.32 (3 H, d, J 7, CHMe); δ_c(CDCl₃): 13.2 (CHMe), 30.9 (CH₂CH₂Ph), 39.3 (CH₂CH₂Ph), 42.3 (CHMe), 53.9 (CHPhS), 65.6 (CH₂OH), 74.3 [C(PhS)₂], 125.7, 126.9, 127.8, 128.0, 128.1, 128.2, 128.5, 128.6, 128.9, 129.0, 129.1, 129.2, 131.2, 131.4, 132.9, 133.2, 136.5, 136.6, 136.8, 137.0 (ArC), 132.0, 132.4, 135.5 and 142.4 (ArC-S, ArC-C); m/z 406 (0.5%, M⁺ – PhSH), 389 (0.2, M^+ – PhSH–OH), 297 (30, M^+ – PhSH – PhS), $189(30, M^+ - PhS - PhS - PhS), 153(35, C_8H_9OS), 135(43, M^+ - PhS), 153(35, C_8H_9OS), 135(43, M^+ - PhS))$ C₈H₈S), 100 (100, PhSH), 109 (33, PhS) and 91 (41, PhCH₂) (Found: $M^+ - C_6H_5S$, 406.1425. $C_{25}H_{26}OS_2$ requires $M - C_6H_5S$) C₆H₅S, 406.1419).

(2RS,3RS)- and (2SR,3RS)-3-Methyl-2,4,4-tris(phenylthio)hexanal 28b. Typical Procedure.—DMSO (0.435 cm³, 6 mmol) in dry dichloromethane (5 cm³) was slowly added to a solution of oxalyl chloride (0.26 cm³, 3 mmol) in dry dichloromethane (15 cm³) at -78 °C. Stirring was continued at this temperature for 10 min, followed by dropwise addition of a solution of the syn-alcohol 27b (1.2 g, 2.72 mmol) in dichloromethane (5 cm³). The reaction mixture was stirred for 20 min and triethylamine (1.9 cm³, 13.6 mmol) was slowly added. Stirring was continued at this temperature for 15 min after which water (13 cm³) was added to the mixture and the cooling bath removed; the mixture was then allowed to warm to room temperature. The organic phase was separated and the aqueous layer re-extracted with dichloromethane. The combined organic extracts were washed successively with dilute hydrochloric acid (10 cm³), water (10 cm³), dilute aqueous sodium carbonate (10 cm^3) and water (10 cm^3) , dried (MgSO₄) and evaporated to dryness to give a crude product. Purification by flash column chromatography on silica eluting with hexanedichloromethane (1:1) gave an 8.1:1 mixture of the aldehydes syn- and anti-28b (836 mg, 70%), as an oil, R_F [hexanedichloromethane (1:1)] 0.25; $v_{max}(film)/cm^{-1}$ 1700 (CO) and 1570 (PhS); $\delta_{\rm H}$ (CDCl₃) (* refers to syn-28b; ** refers to anti-28b) 9.83**, 9.46* (1 H, 2 d, J 3.3 and 6.2, CHO), 7.63-7.18 (15 H, m, PhS), 4.49*, 4.14** (1 H, 2 dd, J 3.6 and 6.2, J 5 and 3.3, CHSPh), 2.88-2.77 (CHMe), 1.79-1.63 (1 H, m, CH_AH_BMe), 1.60-1.51 (1 H, m, CH_AH_BMe), 1.33 (3 H, d, J7, CHMe) and 1.03 (3 H, t, J 7.2, CH₂Me); δ_C syn-28b (CDCl₃) 9.38 (CH₂Me), 14.0 (CHMe), 31.0 (CH₂), 39.1 (CHMe), 61.0 (CHSPh), 74.1 [C(SPh)₂], 127.9, 128.5, 128.6, 129.3, 129.2, 129.3, 132.4, 137.1, 137.3, 137.7 (ArC), 131.5 (ArC-S) and 191.6 (CHO); m/z 329 PhSCHCHO), 149 (CH₃CH₂CHSPh), 123 (23, PhSCH₂), 110 (100, PhSH) and 109 (62, PhS) (Found: $M^+ - C_6H_5S$, 320.1036. $C_{19}H_{21}OS_2$ requires $M - C_6H_5S$, 329.1029).

(2RS,3RS)- and (2SR,3RS)-3-Methyl-2,4,4-tris(phenylthio)heptanal **28d**.—In the same way, the syn-alcohol **27d** (82 mg, 0.18 mmol), oxalyl chloride (0.018 cm³, 0.21 mmol), DMSO (0.031 cm³, 0.44 mmol), triethylamine (0.125 cm³, 0.9 mmol) and water (1 cm³) in the required amounts of dry dichloromethane gave a 4.5:1 mixture of the aldehydes syn- and anti-**28d** (55 mg, 67%), as an oil $R_{\rm F}$ [hexane-dichloromethane

(1:1)] 0.30; $v_{max}(film)/cm^{-1}$ 1700 (CO) and 1570 (PhS); $\delta_{\rm H}({\rm CDCl}_3)$: [* refers to syn-28d; ** refers to anti-28d] 9.80**, 9.43* (1 H, 2 d, J 6.1 and 3.2, CHO), 7.66–7.15 (15 H, m, PhS), 4.48*, 4.35** (1 H, 2 dd, J 3.2 and 4.7, and J 2.5 and 6, CHSPh), 2.81*, 2.75** (1 H, 2 dq, J 4.7 and 6.8, and J 2.5 and 7.2, CHMe) 1.75-1.34 [4 H, m, (CH₂)₂Me], 1.50**, 1.30* (3 H, 2 d, J 7.2 and 6.8, CHMe), 0.71* and 0.70** (3 H, 2 t, J 5.8 and 7, CH₂Me); δ_{C} (CDCl₃) 11.8**, 13.8* (CHMe), 13.7*, 14.1* (CH₂Me), 17.6**, 17.8* (CH₂CH₂Me), 39.5*, 46.4** (CHMe), 39.7**, 4.2* (CH₂CH₂Me), 58.6**, 60.9* (CHPhS), 73.1**, 73.6* [C(PhS)₂], 127.5, 128.5, 128.6, 128.7, 129.2, 129.3, 129.4, 132.3, 133.4, 136.8, 137.1, 137.2, 137.6 (ArC), 130.5, 131.5, 131.6, 132.2, 132.5, 133.2 (ArC-S), 191.6* and 192.0** (CHO); m/z 343 (11%, M⁺ – PhS), 233 (52, M⁺ – PhSH – PhS), 123 $(33, M^+ - PhSH - PhSH - PhS)$, 110 (100, PhSH) and 109 (35, PhS) (Found: $M^+ - C_6H_5S$, 343.1191. $C_{20}H_{23}OS_2$ requires $M - C_6 H_5 S$, 343.1185).

(2RS,3RS)- and (2SR,3RS)-3-Methyl-1-phenyl-2,4,4-tris-(phenylthio)hexanal 28g.—In the same way, the syn-alcohol 27g (1.7 g, 3.3 mmol), oxalyl chloride (0.315 cm³, 3.6 mmol), DMSO (0.515 cm³, 7.25 mmol), triethylamine (2.5 cm³, 18 mmol) and water (19 cm³) in the required amounts of dry dichloromethane afforded a 5.3:1 mixture of the aaldehydes syn- and anti-28g (1.4 g, 83%), as an oil $R_{\rm F}$ [hexane-dichloromethane (1:1)] 0.30; $v_{max}(film)/cm^{-1}$ 1700 (CO) and 1570 (PhS); $\delta_{H}(CDCl_{3})$: [* refers to syn-28g; ** refers to anti-28g] 9.85**, 9.49* (1 H, 2 d, J 6 and 2.9, CHO), 7.79-6.73 (20 H, m, PhS and Ph), 4.63*, 4.35** (1 H, 2 dd, J 2.9 and 4.4, J 2.5 and 6, CHSPh), 3.01-2.73 (3 H, m, CH₂CH₂Ph and CHMe), 2.24–1.83 (2 H, m, CH₂CH₂Ph), 1.57**, 1.38* (3 H, 2 d, J 7.2 and 6.9, CHMe); δ_C(CDCl₃): (syn-28g) 14.1 (CHMe), 30.9 (CH₂CH₂Ph), 39.5 (CHMe), 39.9 (CH₂CH₂Ph), 60.8 (CHPhS), 73.2 [C(PhS)₂], 125.9, 127.9, 128.1, 128.3, 128.7, 128.8, 129.3, 132.2, 136.7, 136.8 (ArC), 131.4, 131.5, 132.3, 141.1 (ArC-S, ArC-C) and 191.7 (CHO); m/z (syn-28g) 405 (4.5%, M⁺ – PhS), 295 (30, M⁺ – PhSH – PhS), 110 (90, PhSH), 109 (33, PhS) and 91 (100, PhCH₂) (Found: $M^+ - C_6 H_6 S$, 405.1355. $C_{25} N_{25} OS_2$ requires M -C₆H₆S, 405.1341).

(2SR,3RS,4RS)- and (2RS,3SR,4RS)-4-Methyl-3,5,5-tris-(phenylthio)heptan-2-ol 29b. Typical Procedure.—An 11:1 mixture of the syn- and anti-aldehydes 28b (745 mg, 1.70 mmol) dissolved in a mixture of hexane (3 cm^3) and ether (2 cm^3) was added dropwise to a solution of methyllithium (1.4 mol dm⁻³ solution in ethyl ether; 1.3 cm³, 1.8 mmol) in hexane (1.3 cm³) kept at -78 °C under argon. After 45 min at this temperature, the cooling bath was removed, the mixture allowed to warm to room temperature (ca. 30 min) and stirring continued for 15 min; water (10 cm³) was then added to the mixture. The organic phase was separated and the aqueous layer re-extracted with ether $(2 \times 10 \text{ cm}^3)$. The combined organic extracts were dried $(MgSO_4)$ and evaporated to dryness. The crude product was purified by flash column chromatography on silica gel eluting with dichloromethane to give an 11:1 mixture of the alcohols anti,syn- and anti,anti-**29b** (520 mg, 67%) as an oil R_F (dichloromethane) 0.44; v_{max} (film)/cm⁻¹ 3600–3200 (OH) and 1570 (D19) 1570 (PhS); m/z 344 (11%, M⁺ – PhSH), 235 (30, M⁺ – PhSH - PhS), 167 (58, PhSCHCHOHMe), 166 (25, PhSC-CHOHMe), 149 (45, PhSC=CHMe), 123 (32, PhSCH₂), 110 (PhSH) and 109 (34, PhS) (Found: $M^+ - C_6H_6S$, 344.1294. $C_{20}H_{24}OS_2$ requires M - C_6H_6S , 344.1263). A portion of this 11:1 mixture of alcohols anti,syn- and anti,anti-29b (70 mg, 0.15 mmol) was treated with benzoic anhydride (87 mg, 0.38 mmol) and DMAP (95 mg, 0.77 mmol) in dry dichloromethane (2.5 cm^3) at room temperature under argon for 24 h. The solution was poured into 3 mol dm⁻³ hydrochloric acid (10 cm³), extracted with dichloromethane $(3 \times 10 \text{ cm}^3)$ and the

extracts were dried (MgSO₄) and concentrated under pressure. This 11:1 mixture of esters (2SR,3RS,4RS)and (2RS,3SR,4RS)-4-methyl-3,5,5-tris(phenylthio)-heptan-2-yl benzoates 30b was separated by flash column chromatography on silica eluting with hexane-dichloromethane (1:1) to give the pure ester anti, syn-30b (65 mg, 78%) as an oil R_F [hexanedichloromethane (1:1)] 0.38; $v_{max}(film)/cm^{-1}$ 1705 (CO) and 1575 (PhS); $\delta_{\rm H}$ (CDCl₃) 7.97–7.15 (20 H, m, PhS and Ph), 5.35 (1 H, dq, J 5.8 and 6.2, MeCHOCOPh), 4.46 (1 H, d, J 5.8, CHSPh), 2.60 (1 H, q, J 7.0, CHMe), 1.91-1.80 (1 H, m, CH_AH_BMe), 1.64–1.52 (1H, m, CH_AH_BMe), 1.42 (3 H, d, J 6.2, MeCHOCOPh), 1.31 (3 H, d, J 7.0, CHMe) and 1.04 (3 H, t J 7.3, CH₂Me); $\delta_{\rm C}$ (CDCl₃) 9.5 (CH₂Me), 13.3 (CHMe), 17.8 (MeCHOCOPh), 30.4 (CH₂), 40.7 (CHMe), 54.1 (CHSPh), 74.9 (MeCHOCOPh), 75.4 [C(SPh)₂], 126.4, 128.2, 128.4, 128.5, 128.8, 129.1, 129.2, 129.8, 129.9, 130.4, 132.7, 137.3 (ArC), 130.3, 131.7, 132.6, 137.0 (ArC-C, ArC-S) and 165.8 $(CO_2Ph); m/z 448 (2\%, M^+ - PhSH), 339 (1, M^+ - PhSH - PhSH)$ PhS), 217 (14, M⁺ – PhSH – PhSH – PhCO₂), 110 (PhSH) and 105 (100, PhCO) (Found: $M^+ - C_6 H_6 S$, 448.1538. $C_{27}H_{28}O_2S_2$ requires $M - C_6H_6S$, 448.1538).

(2SR,3RS,4RS)- and (2RS,3SR,4RS)-4-Methyl-3,5,5-tris-(phenylthio)octan-2-ol 29d.-In the same way, a 1.65:1 mixture of syn- and anti-aldehydes 28d (142 mg, 0.31 mmol), dissolved in a mixture of hexane (0.6 cm^3) and ether (0.4 cm^3) , and methyllithium (1.4 mol dm⁻³ solution in ether; 0.24 cm³, 0.33 mmol) gave a crude product purified by flash column chromatography on silica eluting with dichloromethane to give a 2.6:1 mixture of the alcohols anti,syn- and anti-anti-29d (76 mg, 52%) as an oil $R_{\rm F}$ (dichloromethane) 0.38; $v_{\rm max}$ (film)/cm⁻¹ 3600-3200 (OH) and 1570 (PhS); $\delta_{\rm H}({\rm CDCl}_3)$ (* refers to the major isomer anti,syn-29d and ** refers to the minor isomer anti,anti-29d) 7.72-7.17 (15 H, m, PhS), 5.07**, 3.90-3.86* (1 H, q, J 6.4, and m, CHOH), 4.13*, 3.96** (1 H, 2 d, J 4.1 and 1.9, CHSPh), 2.66**, 2.60* (1 H, 2 dq, J 1.9 and 7, and J 1 and 7, CHMe), 2.43-2.00 (1 H, br s, OH), 1.78-1.40 [1 H, m, (CH₂)₂Me], 1.37**, 1.25* (3 H, 2 d, J7 and 7, CHMe), 1.33**, 1.18* (3 H, 2 d, J 6.4 and 6.4, HOCHMe) and 1.04 (3 H, t, J 7.3, CH_2Me ; $\delta_C(CDCl_3)$ 12.1**, 13.7**, 14.0*, 14.2* (CH_2Me and CHMe), 17.3**, 17.9* (MeCHO), 20.6*, 23.9** (CH₂Me), 39.4*, 39.9** (CH₂CH₂Me), 40.4*, 47.5** (CHMe), 58.4*, 59.3** (CHSPh), 67.3**, 70.7* (MeCOH), 74.6**, 75.0* [C(SPh)₂], 126.3, 126.6, 127.1, 127.4, 128.3, 128.4, 128.7, 128.9, 129.0, 129.2, 129.3, 131.2, 132.0, 136.0, 136.3, 136.7, 136.8, 137.1, 137.2, 137.4 (ArC), 130.2, 131.6, 131.7, 132.2 and 132.9 $(ArC-S); m/z 358 (1\%, M^+ - PhSH), 249 (23, M^+ - PhSH - PhSH)$ PhS), 166 (43, MeCHOHCSPh), 149 (25, MeCHCSPh), 110 (100, PhSH) and 109 (42, PhS) (Found: $M^+ - C_6H_6S$, 358.1418. $M - C_6 H_6 S$ requires 358.1419).

(2SR,3RS,4RS)- and (2RS,3SR,4RS)-4-Methyl-7-phenyl-3,5,5-tris(phenylthio)heptan-2-ol 29g.—In the same way, a 5.3:1 mixture of syn- and anti-aldehydes 28g (1.35 g, 2.62 mmol) dissolved in hexane (5 cm³) and ether (3 cm³), and methyllithium (1.4 mol dm⁻³ solution in ether; 2 cm³, 2.8 mmol) gave after purification by flash column chromatography on silica eluting with dichloromethane the alcohols anti, syn-29g (0.8 g, 57%) as an oil $R_{\rm F}$ (dichloromethane) 0.46; $\nu_{\rm max}$ (film)/cm⁻¹ 3620– 3100 (OH) and 1575 (PhS); $\delta_{\rm H}$ (CDCl₃) 7.81–6.85 (20 H, m, PhS and Ph), 4.19 (1 H, d, J 4.1, CHSPh), 3.91 (1 H, dq, J 4.1 and 6.3, MeCHOH), 3.03 (1 H, dt, J 4 and 13.3, CH₂CH_AH_BPh), 2.76 (1 H, dt, J 4.8 and 13, CH₂CH_AH_BPh), 2.71 (1 H, q, J 7.2, CHMe), 2.16 (2 H, dt, J 4.8 and 13.5, CH_AH_BCH₂Ph and OH), 1.90 (1 H, dt, J 4 and 13.7, CH_AH_BCH₂Ph), 1.34 (3 H, d, J 7.2, CHMe) and 1.21 (3 H, d, J 6.3, MeCHOH); $\delta_{\rm C}$ (CDCl₃) 14.3 (CHMe), 19.8 (MeCHOH), 30.8 (CH₂CH₂Ph), 38.9 (CH₂-CH₂Ph), 40.4 (CHMe), 58.1 (CHSPh), 70.7 (HOCHMe), 74.3

[C(SPh)₂], 125.8, 126.7, 128.3, 128.5, 128.6, 128.8, 129.1, 129.2, 136.0, 136.6, 136.8 (ArC), 131.1, 131.8, 132.7 and 141.4 (ArC-S and ArC-C); m/z 420 (2%, M⁺ – PhSH), 311 (46, M⁺ PhSH - PhS), 201 (32, M⁺ - PhSH - PhSH - PhS), 167 (61, PhSCHCHMeOH), 149 (37, PhSC=CHMe), 123 (32, PhSCH₂), 110 (100, PhSH) and 109 (53, PhS) (Found: M⁺ C_6H_6 , 420.1590. $C_{26}H_{28}OS_2$ requires $M - C_6H_6S$, 420.1590) and a 2.7:1 mixture of the alcohols anti, syn- and anti, anti-29g (0.220 g, 16%), as an oil, $R_{\rm F}$ (dichloromethane) 0.46 and 0.42, respectively. A portion of the alcohol anti, syn-29g (107 mg, 0.20 mmol) was treated with benzoic anhydride (114 mg, 0.5 mmol) and DMAP (125 mg, 1.02 mmol) in dry dichloromethane (3 cm³) as above to give the ester (2SR,3RS,4RS)-4-methyl-7phenyl-3,5,5-tris(phenylthio)heptan-2-yl benzoate anti,syn-30g (93 mg, 74%) as a white solid; m.p. 104-105 °C (from hexane); $R_{\rm F}$ [hexane-dichloromethane (1:1)] 0.40; $\nu_{\rm max}$ (film)/cm⁻¹ 1705 (CO) and 1570 (PhS); $\delta_{\rm H}$ (CDCl₃) 7.82–6.84 (25 H, m, PhS and Ph), 5.34 (1 H, dq, J 6 and 6.3, MeCHOCOPh), 4.46 (1 H, d, J 6, CHSPh), 2.97 (1 H, dt, J 3.8 and 13.8, CH₂CH_AH_BPh), 2.75 $(1 \text{ H}, \text{dt}, J 5 \text{ and } 13.8, \text{CH}_{A}\text{H}_{B}\text{CH}_{2}\text{Ph}), 2.67 (1 \text{ H}, q, J 6, \text{CHMe}),$ 2.26 (2 H, dt, J 5 and 13.8, CH_AH_BCH₂Ph and OH), 1.96 (1 H, dt, J 3.8 and 13.8, CH₂CH_AH_BPh), 1.41 (3 H, d, J 7.2, MeCHOCOPh) and 1.37 (3 H, d, J 6.3, CHMe); $\delta_{\rm C}$ (CDCl₃) 13.5 (CHMe), 179 (MeCHOCOPh), 31.0 (CH₂CH₂Ph), 39.3 (CH₂CH₂Ph), 41.5 (CHMe), 55.1 (CHSPh), 74.1 [C(SPh)₂], 74.8 (MeCHO), 125.9, 126.5, 128.2, 128.4, 128.6, 128.7, 129.0, 129.1, 129.2, 129.8, 130.9, 132.8, 136.7, 136.9 (ArC), 130.3, 131.7, 132.6, 136.8, 141.4 (ArC-S and ArC-C) and 165.8 (CO₂Ph); m/z 524 (0.3%, M^+ – PhS), 415 (37, M^+ – PhS – PhS), 293 (88, M⁺ - PhS - PhS - PhCO₂), 110 (86, PhSH) and 105 (100, PhCO) (Found: $M^+ - C_6 H_5 S$, 524.1352. $C_{33} H_{32} O_2 S_2$ requires $M - C_6 H_5 S$, 524.1836).

Rearrangement of (2SR,3RS,4RS)- and (2RS,3SR,4RS)-4-Methyl-3,5,5-tris(phenylthio)heptan-2-ol 29b.—By method B, an 11:1 mixture of anti, syn- and anti, anti-alcohols 29b (197 mg, 0.43 mmol), triethylamine (0.38 cm^3) and thionyl chloride (0.054 cm^3) in carbon tetrachloride (6 cm³) gave, after flash column chromatography on silica eluting with hexane-dichloromethane (2:1), a 2.4:1 mixture of the vinyl sulfides (E)- and (Z)-(4RS,5RS,6SR)-4-methyl-3,5,6-tris(phenylthio)hept-2-ene 31b (117 mg, 53%), as an oil R_F [hexane-CH₂Cl₂ (2:1)] 0.48; $v_{\text{max}}(\text{film})/\text{cm}^{-1}$ 1570 (PhS); $\delta_{\text{H}}(\text{CDCl}_3)$ 7.57–7.17 (15 H, m, PhS), 6.18 (Z), 5.54 (E) (1 H, q, J 6.9 and 7.1, CH=CSPh), 3.82-3.78 (Z), 3.62-3.59 (E) (2 H, m, CHMe and CHSPh), 3.21 (Z), 3.17 (E)(1 H, dq, J0.7 and 6.8, and m, PhSCHMe), 1.85(E), 1.75(Z)(3 H, 2 d, J 6.9 and 7.1, MeCH=CH₂), 1.40 (E), 1.39 (Z) (3 H, 2 d, J 6.7 and 6.7, CHMe), 1.28 (Z) and 1.21 (E) (3 H, 2 d, J 6.8 and J 6.8, PhSCHMe); $\delta_{C}(CDCl_{3})$ 14.7 (E), 16.1 (Z) (CHMe), 16.7 (E), 17.6 (Z) (MeCH=CSPh), 18.3 (Z), 19.1 (E) (PhSCHMe), 41.5 (E), 45.7 (Z) (CHMe), 48.1 (E), 48.1 (Z) (PhSCHMe), 60.7 (E), 61.0 (Z) (CHSPh), 125.7, 126.0, 126.3, 126.5, 126.9, 127.2, 127.5, 127.6, 128.3, 128.5, 128.8, 128.9, 129.0, 129.1, 130.8, 130.9, 131.4, 131.9, 132.4, 132.5, 133.8 (ArC, MeCH=CSPh), 134.7, 135.5, 137.8 and 138.5 (ArC-S, MeCH=*C*SPh); m/z 436 (0.8%, M⁺), 327 (58, M⁺ – PhS), 217 (26, M⁺ – PhS – PhSH), 149 (100, MeCHCSPh), 137 (53, MeCHSPh), 110 (75, PhSH) and 109 (74, PhS) (Found: M⁺, 436.1329. C₂₆H₂₈S₃ requires M, 436.1347).

Rearrangement of (2SR,3RS,4RS)- and (2RS,3SR,4RS)-4-Methyl-3,5,5-tris(phenylthio)octan-2-ol **29d**.—By method B, a 2.6:1 mixture of anti,syn- and anti,anti-alcohols **29d** (32 mg, 0.07 mmol), triethylamine (0.057 cm³) and thionyl chloride (0.009 cm³) in carbon tetrachloride (1 cm³) gave, after flash column chromatography on silica eluting with hexanedichloromethane (2:1), a 2.7:1 mixture of the vinyl sulfides (E)and (Z)-(4RS,5RS,6SR)-4-methyl-3,5,6-tris(phenylthio)oct-2ene **31d** (17 mg, 55%) as an oil $R_{\rm F}$ [hexane–dichloromethane (2:1)] 0.50; $\nu_{\rm max}({\rm film})/{\rm cm}^{-1}$ 1570 (PhS); $\delta_{\rm H}({\rm CDCl}_3)$ 7.52–7.17 (15 H, m, PhS), 6.04 (Z), 5.48 (Z) (1 H, 2 t, J 7 and 7.6, CH=CSPh), 3.78–3.69 (E), 3.57–3.52 (Z) (2 H, m, CHMe and CHSPh), 3.20–3.13 (E), 2.98–2.92 (Z) (1 H, m, PhSCHMe), 2.35–2.08 (2 H, m, MeCH₂CH=CSPh), 1.36 (Z), 1.35 (E) (3 H, 2 d, J 6.5 and 6.7, CHMe), 1.25 (Z) and 1.15 (E) (C H, 2 d, J 6.7 and 6.8, PhSCHMe); m/z 450 (2%, M⁺), 341 (100, M⁺ – PhS), 231 (28, M⁺ – PhS – PhSH), 149 (86, MeCHCSPh), 137 (46 MeCHSPh), 110 (42, PhSH) and 109 (30, PhS) (Found: M⁺, 450.1503. C₂₇H₃₀S₃ requires M, 450.1503).

Rearrangement of (2SR,3RS,4RS)-4-Methyl-7-phenyl-3,5,5tris(phenylthio)heptan-2-ol 29g.-By method B, the anti,synalcohol 29g (132 mg, 0.25 mmol), triethylamine (0.2 cm³) and thionyl chloride (0.030 cm³) in carbon tetrachloride (6 cm³) gave, after flash column chromatography on silica eluting with hexane-dichloromethane (2:1), a 4:1 mixture of the vinyl sulfides (E)- and (Z)-(4RS,5RS,6SR)-4-methyl-1-phenyl-3,5,6tris(phenylthio)hept-2-ene **31g** (76 mg, 60%), as an oil $R_{\rm F}$ [hexane-dichloromethane (2:1)] 0.46; $v_{max}(film)/cm^{-1}$ 1570 (PhS); δ_H(CDCl₃) 7.50–6.87 (20 H, m, PhS and Ph), 6.19 (Z), 5.55(E)(1 H, dd, J6.5 and 7.8, and t, J7.4, CH=CSPh), 3.81(Z), 3.80 (E) (1 H, 2 dd, J 7.8 and 15.6, and J 7.8 and 15.6, CH_AH_BCH=CSPh), 3.60 (1 H, d, J 6.5, CHSPh), 3.51 (1 H, dq, J 2.3 and 6.4, CHMe), 3.41 (Z), 3.39 (E) (1 H, 2 dd, J 6.5 and 15.6, $CH_AH_BCH=CSPh$), 3.30(E), 3.02(Z)(1H, m, PhSCHMe), 1.36 (Z), 1.33 (E) (3 H, 2 d, 56.7 and 6.7, CHMe) and 1.15 (3 H, d, J 6.8, PhSCHMe); δ_{c} (CDCl₃) 16.7 (E), 17.2 (Z), 18.6 (Z), 19.3 (E) (CHMe and PhSCHMe), 34.7 (E), 36.5 (Z) (CH₂CH₂Ph), 41.7 (E), 45.0 (Z) (CHMe), 47.9 (E), 48.3 (Z) (PhSCHMe), 60.7 (E), 60.8 (Z) (CHSPh), 126.0, 126.1, 126.1, 126.2, 126.3, 126.9, 127.5, 128.1, 128.4, 128.5, 128.7, 128.8, 129.0, 129.1, 130.7, 130.8, 132.3, 132.4, 132.5, 132.8, 137.3 (ArC, MeCH=CSPh), 135.5, 136.6, 137.7, 138.9 and 139.9 (ArC-S, MeCH=CSPh); m/z 403 (7%, M⁺ – PhS), 149 (58, MeCHC-SPh), 137 (38, MeCHSPh), 110 (100, PhSH), 109 (63, PhS) and 91 (98, PhCH₂) (Found: $M^+ - C_6H_5S$, 403.1537. $M - C_6H_5S$ requires 403.1548).

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References

- 1 P. Blatcher and S. Warren, J. Chem. Soc., Perkin Trans 1, 1985, 1055.
- 2 (a) [1,2] PhS shift: P. Brownbridge and S. Warren, J. Chem. Soc., Perkin Trans. 1, 1977, 1131, 2272; (b) [1,3]-PhS shift: P. Brownbridge and S. Warren, J. Chem. Soc., Perkin Trans. 1, 1976, 2125.
- 3 M.-J. Villa and S. Warren, Tetrahedron Lett., 1989, 30, 5933
- 4 P. de Mayo, Rearrangements in Ground and Excited States, Academic Press, New York, 1980.
- 5 S. M. Tuladhar and A. G. Fallis, Tetrahedron Lett., 1987, 28, 523.
- 6 M.-J. Villa and S. Warren, J. Chem. Soc., Perkin Trans. 1, 1994, 1569.
- 7 S. Yamagiwa, N. Hoshi, H. Sato, H. Kosugi and H. Uda, J. Chem. Soc., Perkin Trans. 1, 1978, 214.
- 8 M. N. Paddon-Row, N. G. Rondan and K. N. Houk, J. Am. Chem. Soc., 1982, 104, 7162.
- 9 (a) K. Tomioka, H. Kawasaki and K. Koga, *Tetrahedron Lett.*, 1985,
 26, 3027; (b) H. Kawasaki, K. Tomioka and K. Koga, *Tetrahedron Lett.*, 1985,
 26, 3031; (c) K. Tomioka, M. Sudani, Y. Shinmi and K. Koga, *Chem. Lett.*, 1985, 329.
- (a) I. Fleming, J. H. M. Hill, D. Parker and D. Waterson, J. Chem. Soc., Chem. Commun., 1985, 318; (b) I. Fleming and J. J. Lewis, J. Chem. Soc., Chem. Commun., 1985, 149; (c) H.-F. Chow and I. Fleming, Tetrahedron Lett., 1985, 25, 397; (d) W. Bernhard, I. Fleming and D. Waterson, J. Chem. Soc., Chem. Commun., 1984, 28; (e) Y. Yamamoto and K. Maruyama, J. Chem. Soc., Chem. Commun., 1984, 904.

- 11 (a) B. Capon and S. P. McManus in Neighbouring Group Participation, 1976, Plenum Press, New York; (b) R. Bird and C. J. M. Stirling, J. Chem. Soc., Perkin Trans. 2, 1973, 1221; (c) E. Block in Reactions of Organosulfur Compounds, 1978, Academic Press, New York.
- 12 (a) P. Brownbridge, I. Fleming, A. Pearce and S. Warren, J. Chem. Soc., Chem. Commun., 1976, 751; (b) I. Fleming, I. Paterson and A. Pearce, J. Chem. Soc., Perkin Trans. 1, 1981, 256.
- 13 K. Omura and D. Swern, Tetrahedron, 1978, 34, 1651.
- 14 E. J. Corey and G. Schmidt, *Tetrahedron Lett.*, 1979, 399 and references cited therein.
- 15 K. Chibale and S. Warren, Tetrahedron Lett., 1994, 35, 3991.
- 16 V. K. Aggarwal, I. Coldham, S. McIntyre, F. H. Sansbury, M.-J. Villa
- and S. Warren, Tetrahedron Lett., 1988, 29, 4885.
- 17 J. Eames, S. McIntyre and S. Warren, unpublished results.
- 18 S. K. Armstrong, E. W. Collington and S. Warren, J. Chem. Soc., Perkin Trans. 1, 1994, 515.

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